

## Periodic Table Named-Groups Worksheet – 2

**DIRECTIONS:** Use the Periodic Table shown below to answer each question. using the letters in each "box" to identify the required element(s).

The diagram shows a crossword puzzle grid with the following letters placed in specific cells:

- Row 1:** A (1,1), B (15,1)
- Row 2:** C (1,2), D (10,2), E (12,2)
- Row 3:** F (2,3), G (7,3), H (9,3), K (15,3)
- Row 4:** L (2,4), I (12,4)
- Row 5:** J (1,5), Z (3,5), M (7,5), N (10,5), O (13,5), P (15,5)
- Row 6:** Q (1,6), R (3,6), S (7,6), T (8,6), U (9,6)

Below the main grid, there is a separate row of 15 cells. The first cell contains 'V' and the second cell contains 'X'. Arrows point from the letter 'R' in the main grid to these two cells.

The remaining cells in the separate row are empty, with the letter 'W' in the 10th cell and 'Y' in the 11th cell.

List elements in alphabetical order when more than 1 element answers a question.

- \_\_\_\_\_ 1) Element **A** is in a box all by itself because: **A)** it is really a Noble Gas **B)** it is a gas **C)** it is NOT a gas **D)** it does NOT share the properties of the other elements in its group **E)** it is a liquid
- \_\_\_\_\_ 2) Which of these elements is/are highly reactive? **A)** "C" **B)** "E" **C)** "K" **D)** A, B, and C **E)** A and B only
- \_\_\_\_\_ 3) What is the letter of the element whose atomic number is closest to the element labeled "Z"?
- \_\_\_\_\_ 4) Which letter labels the halogen element with the lowest atomic number?
- \_\_\_\_\_ 5) Which of these elements is the least reactive? **A)** "J" **B)** "Y" **C)** "A" **D)** "O" **E)** "B"
- \_\_\_\_\_ 6) Which of these elements is/are NOT transition elements/metals. (list ALL that are not transition elements)? **A)** "C" **B)** "N" **C)** "K" **D)** "N" **E)** "Y"

continued on other side

# Periodic Tables Answer Key

**O García**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

## Periodic Tables Answer Key:

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