

Name: _____

Period: _____

Periodic Table WebQuest

Instructions: In each of the websites listed, find the information asked for.



Part I: "Get Organized Periodically"

<http://www.chemicalsciences.com/lesson/periodic.html>

1. Why are the elements placed in specific places on the Periodic Table?
2. Periods are _____ that run from left to right.
3. Elements in the same period have the same _____.
4. Every element in the first period has _____ shell for its _____. Every element in the second period has _____ for its _____. See the pattern?
5. Groups are _____ that run from top to bottom.
6. The elements of a group have the same number of _____ in their _____ shell.
7. Every element in group one has _____ electron in its outer shell. Every element in group two has _____ electrons in its outer shell.
8. Hydrogen is special because it can act like two groups, _____ and _____.
9. Hydrogen sometimes is _____ an electron and sometimes it has an _____ electron.
10. Although helium has only _____ electrons in its outer shell, it is grouped with elements that have _____.
11. The green elements on this table are called _____ elements. They each have two electrons in their outer shell.

Part II: "Family Fun" Go to <http://chemicalsciences.com/>

12. Click on Alkali Metals (left bar) and answer the following questions.
 - a. What is the group number? _____
 - b. Are these metals reactive? _____
 - c. Do these metals occur freely in nature? _____
 - d. How many electrons are in their outer shell? _____

Periodic Table Webquest Answer Key

Karin Nielsen-Saines



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Table of Contents Periodic Table Webquest Answer Key

1. Understanding the eBook Periodic Table Webquest Answer Key
 - The Rise of Digital Reading Periodic Table Webquest Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Webquest Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Webquest Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Webquest Answer Key
 - Personalized Recommendations
 - Periodic Table Webquest Answer Key User Reviews and Ratings
 - Periodic Table Webquest Answer Key and Bestseller Lists
5. Accessing Periodic Table Webquest Answer Key Free and Paid eBooks
 - Periodic Table Webquest Answer Key Public Domain eBooks
 - Periodic Table Webquest Answer Key eBook Subscription Services
 - Periodic Table Webquest Answer Key Budget-Friendly Options
6. Navigating Periodic Table Webquest Answer Key eBook Formats

- ePub, PDF, MOBI, and More
- Periodic Table Webquest Answer Key Compatibility with Devices
- Periodic Table Webquest Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table Webquest Answer Key
 - Highlighting and Note-Taking Periodic Table Webquest Answer Key
 - Interactive Elements Periodic Table Webquest Answer Key
- 8. Staying Engaged with Periodic Table Webquest Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Webquest Answer Key
- 9. Balancing eBooks and Physical Books Periodic Table Webquest Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Table Webquest Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Table Webquest Answer Key
 - Setting Reading Goals Periodic Table Webquest Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Table Webquest Answer Key
 - Fact-Checking eBook Content of Periodic Table Webquest Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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