

# Periodic Table Basics

**Step 1:** Complete the square for each element by filling in the atomic number, name, & atomic mass.

**Step 2:** Determine the number of protons, neutrons, and electrons in an atom of each element.

**Step 3:** Identify if the element is a solid, liquid, or gas at room temperature.

**Step 4:** Give the melting (M.P.) and boiling points (B.P.) in degrees Celsius.

**Step 5:** List at least three physical or chemical properties for each element.

**Step 6:** List at least three uses for each element.

**Step 7:** Draw a Bohr diagram and Lewis Structure to show the arrangement of electrons and the number of valence electrons.

**Step 8:** Use the following colors to shade in the square for each element. You should **ONLY** color the small square in the upper left-hand corner and not the entire card.

Green = Li & Na  
Orange = B & Al

Pink = O & S  
Red = C & Si

Blue = Be & Mg  
Tan = N & P

Purple = F & Cl  
Yellow = He, Ne, & Ar

**Step 9:** Cut the cards apart and arrange according to atomic number in the pattern shown. Once you have the cards arranged in the correct order, glue them to a large sheet of construction paper.

1							2
3	4	5	6	7	8	9	10
11	12	13	14	15	16	17	18

Done? Answer the questions on the worksheet using the information on your Periodic Table!

**The  
Science  
Spot**

Need information? Visit the Periodic Table links on the Chemistry page of the Kid Zone!

Go to <http://sciencespot.net/>  
and click the Kid Zone graphic!

① <div style="text-align: center; font-size: 2em;">B</div>	P= ② N=____ E=____	③ O S O L O G
	M.P. =____ B.P. =____	
Properties ④		
Uses ⑤		
Bohr Diagram    Lewis Structure		
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# Periodic Table Basics Answers The School Spot

**Ray Petty**



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## **Table of Contents Periodic Table Basics Answers The School Spot**

1. Understanding the eBook Periodic Table Basics Answers The School Spot
  - The Rise of Digital Reading Periodic Table Basics Answers The School Spot
  - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Basics Answers The School Spot
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Periodic Table Basics Answers The School Spot
  - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Basics Answers The School Spot

- Personalized Recommendations
  - Periodic Table Basics Answers The School Spot User Reviews and Ratings
  - Periodic Table Basics Answers The School Spot and Bestseller Lists
5. Accessing Periodic Table Basics Answers The School Spot Free and Paid eBooks
- Periodic Table Basics Answers The School Spot Public Domain eBooks
  - Periodic Table Basics Answers The School Spot eBook Subscription Services
  - Periodic Table Basics Answers The School Spot Budget-Friendly Options
6. Navigating Periodic Table Basics Answers The School Spot eBook Formats
- ePub, PDF, MOBI, and More
  - Periodic Table Basics Answers The School Spot Compatibility with Devices
  - Periodic Table Basics Answers The School Spot Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Periodic Table Basics Answers The School Spot
  - Highlighting and Note-Taking Periodic Table Basics Answers The School Spot
  - Interactive Elements Periodic Table Basics Answers The School Spot
8. Staying Engaged with Periodic Table Basics Answers The School Spot
- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Periodic Table Basics Answers The School Spot
9. Balancing eBooks and Physical Books Periodic Table Basics Answers The School Spot
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Periodic Table Basics Answers The School Spot
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Periodic Table Basics Answers The School Spot
- Setting Reading Goals Periodic Table Basics Answers The School Spot
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodic Table Basics Answers The School Spot

- Fact-Checking eBook Content of Periodic Table Basics Answers The School Spot
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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