# PERIODIC CLASSIFICATION OF ELEMENT

# DOBERNIER TRIADS:-

→ Aronged elements in the genoup of 3.
→ Aromic mass of middle cloment is avenuer of atronic mass of other two elements.

F 23 29 } Triad.

#### + OHOLUBOONS

an known elements were not grouped

#### Lother Meyer's Curves-



-> fit hand methods -> top position

\* Ackaline easem metals = descinding \* Halogen = ascending position

## MODERN PERIODIC TABLES

+ long bown of pewlock table

\* Reviodic peroperation are periodic

Probability Repetron of similary

properates of elements after evolutar

configuration

Transition elements; they have incompletely fluid a-orbitals in their ground.

because of Completely Little dqubitale (Pando transition elements)

Representative Clement > combined elements
of s and p brock quie known Qu
supresentative clements

# NEWLAND'S OCTAVE RULE:

then elements are aumanged in increasing audit of atomic man. then every ath element stanting from first with stone some property. By- U Be B C N O F

#### Dynus backs t-

s-voted expte calcium only a tre periode of ineut gade s-nes worked on heavils elements.

#### MENDLEEVS PERIODIC

 Paulodic propulty of elements au peulodic function of that aromic masses

#### Menits

- samplified study - same vacant space was telefour the elements which wave not because at that time

# Eka atimunium + Ga.

corrected aromic mass of element

#### Discount brack &

to Could not expeal periodicity.

#### MODERN PERIODIC TRBLE:

18 groups and Francisco

S-810ck = G1 and G2 = ms<sup>1-2</sup> P-810ck = 915 - 916 = ns<sup>2</sup> np<sup>1-6</sup>

D-8lock = Q s - Q ιz = (n-1) d 1-10 ns 0-2

f βlock = lanthenoid = 2 σ 58 - 71

G3 and Ps

GS and Pa

#### Diagonal Relation:

No mg AL charge ratio

#### Nomenclature;

5 - pent Sugion ium

2 bt 2 scot unnitporting

30 tri 80 inn

#### EFFECTIVE NUCLEAR CHARGE

Tends - Peniod - Zett?

group Zott almost remains

scheeling + s) +>d>t

- more the charge, more is Zets. - more the charge, less In Zets.

#### TONISATION ENERGY

-> Ehbrigg stequired to elemone out Electron gram on isolated gastons altom-

Man - Migs + C - XE1 AHF + WE WANTED - M'Up + C - XE2 COMMANY

#### Trender Zell in IE

Proup - Zeet + = IE+

Gu = G > AL > Ga > In Gu = c > gi > Ge > gb > Sn. 1 < 8 < 6 < C < 0 < N < F < NE

# ATOMIC RADIUS:

of automost show and electrone

\* But its imposible to find exact atoms stading

Retorn -z = arom is to small
are arom doe no have any bunday

\* Ovalent Radius Hall of the Internucleus distance byo 2 atomic bounded by Covalent word-



Merallic Radius + Hall of the

bonded by a metallic bond.



Evanderica at Radius - Half of the Intermetecular distance by A mediate morecuse in non-tonded.



Previor + Zett + - Atomic Radius +
group - Zett + - Atomic Radius +
group - Zett + - Atomic Radius +
Tonic Radius + Rasion < Recutrat Ranco

## ELECTRON GAIN ENTHAUPY:-

an e- vis added to cutermost cubit of an Esciared gaveous atom is known as Electron gain contrally.

Note: In majority cases degts is -we (exometric)

Exception = 8 7 degts = +ve

E-Ception = S

Notes gas Configuration

Notes succession electron gain emmassy is always the

PERPS; DegH of Zegg

Power - Zett + + acquit

GE = S > SE > TE > FO > O PERSON!

\* due to small size presson!

\* d-e: Exeputation

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