

Chemistry: Periodic Table Unit Study Guide

1. Name _____

2. Date _____

3. Class _____

4. Teacher _____

Periodic Table Study Guide

Major Groups and Properties of the Periodic Table

1. What are the major families of elements in the periodic table?

- Alkali metals (Group 1)
- Alkaline earth metals (Group 2)
- Transition metals (Groups 3-10)
- Post-transition metals (Groups 11-16)
- Nonmetals (Groups 17-18)
- Lanthanides and Actinides (f-block)

2. List the following families in the periodic table: Alkali metals, Alkaline earth metals, Transition metals, Post-transition metals, Nonmetals, Lanthanides and Actinides. For each family, write down the main group number and the color of the family.

5. Draw the periodic table with the following families highlighted:

6. Draw the periodic table with the following families highlighted:

7. Draw the periodic table with the following families highlighted:

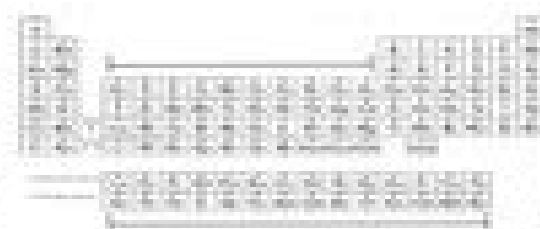
8. Draw the periodic table with the following families highlighted:

9. Draw the periodic table with the following families highlighted:

10. Draw the periodic table with the following families highlighted:

11. Draw the periodic table with the following families highlighted:

12. Draw the periodic table with the following families highlighted:



Alkali metals

Alkaline earth metals

Transition metals

Post-transition metals

Nonmetals

Lanthanides and Actinides

13. Fill in the following table:

Element	Atomic Number	Symbol	Electron Configuration	Group
Hydrogen	1	H	1s ¹	1
Helium	2	He	1s ²	18
Lithium	3	Li	[He] 2s ¹	1
Beryllium	4	Be	[He] 2s ²	2
Boron	5	B	[He] 2s ² 2p ¹	13
Carbon	6	C	[He] 2s ² 2p ²	14
Nitrogen	7	N	[He] 2s ² 2p ³	15
Oxygen	8	O	[He] 2s ² 2p ⁴	16
Fluorine	9	F	[He] 2s ² 2p ⁵	17
Neon	10	Ne	[He] 2s ² 2p ⁶	18
Sodium	11	Na	[Ne] 3s ¹	1
Magnesium	12	Mg	[Ne] 3s ²	2
Aluminum	13	Al	[Ne] 3s ² 3p ¹	13
Silicon	14	Si	[Ne] 3s ² 3p ²	14
Phosphorus	15	P	[Ne] 3s ² 3p ³	15
Sulfur	16	S	[Ne] 3s ² 3p ⁴	16
Chlorine	17	Cl	[Ne] 3s ² 3p ⁵	17
Argon	18	Ar	[Ne] 3s ² 3p ⁶	18
Potassium	19	K	[Ar] 4s ¹	1
Calcium	20	Ca	[Ar] 4s ²	2
Scandium	21	Sc	[Ar] 3d ¹ 4s ²	3
Titanium	22	Ti	[Ar] 3d ² 4s ²	4
Vanadium	23	V	[Ar] 3d ³ 4s ²	5
Chromium	24	Cr	[Ar] 3d ⁵ 4s ¹	6
Manganese	25	Mn	[Ar] 3d ⁵ 4s ²	7
Iron	26	Fe	[Ar] 3d ⁶ 4s ²	8
Cobalt	27	Co	[Ar] 3d ⁷ 4s ²	9
Nickel	28	Ni	[Ar] 3d ⁸ 4s ²	10
Copper	29	Cu	[Ar] 3d ¹⁰ 4s ¹	11
Zinc	30	Zn	[Ar] 3d ¹⁰ 4s ²	12
Gallium	31	Ga	[Ar] 3d ¹⁰ 4s ² 4p ¹	13
Germanium	32	Ge	[Ar] 3d ¹⁰ 4s ² 4p ²	14
Arsenic	33	As	[Ar] 3d ¹⁰ 4s ² 4p ³	15
Selenium	34	Se	[Ar] 3d ¹⁰ 4s ² 4p ⁴	16
Bromine	35	Br	[Ar] 3d ¹⁰ 4s ² 4p ⁵	17
Krypton	36	Kr	[Ar] 3d ¹⁰ 4s ² 4p ⁶	18
Rubidium	37	Rb	[Kr] 5s ¹	1
Strontium	38	Sr	[Kr] 5s ²	2
Yttrium	39	Y	[Kr] 4d ¹ 5s ²	3
Zirconium	40	Zr	[Kr] 4d ² 5s ²	4
Niobium	41	Nb	[Kr] 4d ⁴ 5s ¹	5
Molybdenum	42	Mo	[Kr] 4d ⁵ 5s ¹	6
Technetium	43	Tc	[Kr] 4d ⁵ 5s ²	7
Ruthenium	44	Ru	[Kr] 4d ⁶ 5s ²	8
Rhodium	45	Rh	[Kr] 4d ⁷ 5s ²	9
Palladium	46	Pd	[Kr] 4d ¹⁰ 5s ⁰	10
Silver	47	Ag	[Kr] 4d ¹⁰ 5s ¹	11
Cadmium	48	Cd	[Kr] 4d ¹⁰ 5s ²	12
Indium	49	In	[Kr] 4d ¹⁰ 5s ² 5p ¹	13
Sn	50	Sn	[Kr] 4d ¹⁰ 5s ² 5p ²	14
Sb	51	Sb	[Kr] 4d ¹⁰ 5s ² 5p ³	15
Te	52	Te	[Kr] 4d ¹⁰ 5s ² 5p	

Periodic Table And Chemistry Study Guide Answers

Ensheng Dong



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web the yamaha aerox is a lineup of single cylinder scooters made by yamaha since 1997 available in either 50 cc or 100 cc for the european market and 125 cc or 155 cc for the southeast asian market with several different body designs markets europe slight changes were made in 2003

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web if we talk about yamaha aerox 155 engine specs then the gasoline engine displacement is 155 cc aerox 155 is available with cvt transmission the aerox 155 is a 2 seater scooter and has a length of 1980 mm the width of 700 mm and a wheelbase of 1350 mm along with a ground clearance of 145 mm