

Percent Yield

1. In a certain chemical reaction, 100 g of reactant A (molar mass 100 g/mol) reacts with 100 g of reactant B (molar mass 100 g/mol) to produce 100 g of product C (molar mass 100 g/mol). What is the percent yield of product C?
2. In a certain chemical reaction, 100 g of reactant A (molar mass 100 g/mol) reacts with 100 g of reactant B (molar mass 100 g/mol) to produce 100 g of product C (molar mass 100 g/mol). What is the percent yield of product C?
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7. In a certain chemical reaction, 100 g of reactant A (molar mass 100 g/mol) reacts with 100 g of reactant B (molar mass 100 g/mol) to produce 100 g of product C (molar mass 100 g/mol). What is the percent yield of product C?
8. In a certain chemical reaction, 100 g of reactant A (molar mass 100 g/mol) reacts with 100 g of reactant B (molar mass 100 g/mol) to produce 100 g of product C (molar mass 100 g/mol). What is the percent yield of product C?
9. In a certain chemical reaction, 100 g of reactant A (molar mass 100 g/mol) reacts with 100 g of reactant B (molar mass 100 g/mol) to produce 100 g of product C (molar mass 100 g/mol). What is the percent yield of product C?
10. In a certain chemical reaction, 100 g of reactant A (molar mass 100 g/mol) reacts with 100 g of reactant B (molar mass 100 g/mol) to produce 100 g of product C (molar mass 100 g/mol). What is the percent yield of product C?

Percent Yield Skill Practice 44

Didier Musso



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