

Answers

$Q = mc\Delta T$, where Q = heat energy, m = mass, and ΔT = change in temp.
Remember, $\Delta T = (T_{\text{final}} - T_{\text{initial}})$. Show all work and proper units.

1. A 15.75-g piece of iron absorbs 1086.75 joules of heat energy, and its temperature changes from 25°C to 175°C. Calculate the specific heat capacity of iron.

$$C = \frac{Q}{m(T_f - T_i)} = \frac{1086.75}{15.75(175-25)} = 0.46 \text{ J/g}^\circ\text{C}$$

2. How many joules of heat are needed to raise the temperature of 10.0 g of aluminum from 22°C to 55°C, if the specific heat of aluminum is 0.90 J/g°C?

$$Q = mC(T_f - T_i) = 10.0\text{g} (0.90\text{J/g}^\circ\text{C})(55-22) = 297 \text{ J}$$

3. Calculate the specific heat capacity of a piece of wood if 1500.0 g of the wood absorbs 67,500 joules of heat, and its temperature changes from 32°C to 57°C.

$$C = \frac{Q}{m(T_f - T_i)} = \frac{67500 \text{ J}}{(1500 \text{ g})(57-32)} = 1.8 \text{ J/g}^\circ\text{C}$$

4. 100.0 g of 4.0°C water is heated until its temperature is 37°C. Calculate the amount of heat energy needed to cause this rise in temperature.

$$Q = mC(T_f - T_i) = 100\text{g}(4.184\text{J/g}^\circ\text{C})(37 - 4) = 14000 \text{ J}$$

5. 25.0 g of mercury is heated from 25°C to 155°C, and absorbs 455 joules of heat in the process. Calculate the specific heat capacity of mercury.

$$C = \frac{Q}{m(T_f - T_i)} = \frac{455 \text{ J}}{(25\text{g})(155-25)} = 0.14 \text{ J/g}^\circ\text{C}$$



Physical Science If8767 Heat Calculations

W Nerst

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Teaching Heat Transfer and Heat Exchange Brian Melvin Evenson, 2004

Worked Problems in Heat, Thermodynamics, and Kinetic Theory for Physics Students L. Pincherle, 1966 Worked Problems in Heat Thermodynamics and Kinetic Theory for Physics Students is a complementary to textbooks in physics This book is a collection of exercise problems that have been part of tutorial classes in heat and thermodynamics at the University of London This collection of exercise problems with answers that are fully worked out deals with various topics This book poses problems covering the definition of temperature such as calculating the assigned value of the temperature of boiling water under specific conditions This text also gives example of problems dealing with the first law

The New Heat Theorem Walther Nernst, 1969

An Introduction to Heat Transfer A. J. Ede, 2000

The Heat Equation D. V. Widder, 1976-01-22 The Heat Equation

Calculations in Heat, Light and Sound ... S. M. Niven, 1964

Heat-transfer Calculations by Finite Differences George Merrick Dusenberre, 1961

Heat and Thermodynamics Mark Waldo Zemansky, 1968 Written for general physics courses that emphasize temperature dependent phenomena this text deals with large scale thermal phenomena and then proceeds to small scale less accessible phenomena Examples of calculations are presented after important formulas are derived and actual thermodynamic experiments are explained in detail

Physics Heat Gordon Ball Wilkes, 1926

The Theory of Heat Thomas Preston, 1904

Analysis of Heat Equations on Domains. (LMS-31) El-Maati Ouhabaz, 2004-10-31 This is the first comprehensive reference published on heat equations associated with non self adjoint uniformly elliptic operators The author provides introductory materials for those unfamiliar with the underlying mathematics and background needed to understand the properties of heat equations He then treats L_p properties of solutions to a wide class of heat equations that have been developed over the last fifteen years These primarily concern the interplay of heat equations in functional analysis spectral theory and mathematical physics This book addresses new developments and applications of Gaussian upper bounds to spectral theory In particular it shows how such bounds can be used in order to

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Thermodynamics Michael Sprackling, 1993 This undergraduate text presents the core topics in thermal physics using the problem based learning approach The book has combined the aim of promoting understanding through problem solving and by putting many of the problems in traditional examination form providing exam preparation Thermal Measurements and Inverse Techniques Helcio R.B. Orlande, Olivier Fudym, Denis Maillet, Renato M. Cotta, 2011-05-24 With its uncommon

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Thermal Physics Michael Sprackling, 1991 This is a general exposition of the principles governing the properties and effects of heat Each chapter is followed by examples problems and notes which should enhance the student's understanding of and feel for the subject *Formulas and tables for heating, German formulas and tables* John Henry Kinealy, 1899

Worked Problems in Heat, Thermodynamics and Kinetic Theory L. Pincherle, 1966-10 **The Analytical Theory of Heat** Joseph Fourier, 2007-10-01 In this pioneering work of mathematics Joseph Fourier shows how the conduction of heat in solid bodies can be analyzed in terms of an infinite mathematical series Known as the Fourier Series this was the first correct theory on heat diffusion and continues to be used in present day analysis For anyone interested in the theory of heat or in the mathematical tools developed by Fourier this classic work remains indispensable Born the son of a French tailor JOSEPH FOURIER 1768 1830 was a mathematician Egyptologist and politician whose strong influence on mathematical physics continues to this day His other works include Description of Egypt and Analysis of Determinate Equations *The New Heat Theorem* W Nerst, 1984-06-01

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