

# Periodic Table of the Elements



11.01

H

Hydrogen

3.04

Li

Lithium

6.94

Be

Beryllium

12.01

Na

Sodium

24.31

Mg

Magnesium

14

Si

Silicon

atomic number

atomic weight

symbol

name

black

solid

blue

liquid

red

gas

white

synthetically prepared

near ambient temp

alkali metals

alkaline earth metals

transitional metals

other metals

nonmetals

noble gases

2

He

Helium

5

B

Boron

6

C

Carbon

7

N

Nitrogen

8

O

Oxygen

9

F

Fluorine

10

Ne

Neon

13

Al

Aluminum

14

Si

Silicon

15

P

Phosphorus

16

S

Sulfur

17

Cl

Chlorine

18

Ar

Argon

19

K

Potassium

20

Ca

Calcium

21

Sc

Scandium

22

Ti

Titanium

23

V

Vanadium

24

Cr

Chromium

25

Mn

Manganese

26

Fe

Iron

27

Co

Cobalt

28

Ni

Nickel

29

Cu

Copper

30

Zn

Zinc

31

Ga

Gallium

32

Ge

Germanium

33

As

Arsenic

34

Se

Selenium

35

Br

Bromine

36

Kr

Krypton

37

Rb

Rubidium

38

Sr

Strontium

39

Y

Yttrium

40

Zr

Zirconium

41

Nb

Niobium

42

Mo

Molybdenum

43

Tc

Technetium

44

Ru

Ruthenium

45

Rh

Rhodium

46

Pd

Palladium

47

Ag

Silver

48

Cd

Cadmium

49

In

Indium

50

Sn

Tin

51

Sb

Antimony

52

Te

Tellurium

53

I

Iodine

54

Xe

Xenon

55

Cs

Cesium

56

Ba

Barium

57

La

Lanthanum

58

Ce

Cerium

59

Pr

Praseodymium

60

Nd

Neodymium

61

Pm

Promethium

62

Sm

Samarium

63

Eu

Europium

64

Gd

Gadolinium

65

Tb

Terbium

66

Dy

Dysprosium

67

Ho

Holmium

68

Er

Erbium

69

Tm

Thulium

70

Yb

Ytterbium

71

Lu

Lutetium

72

Hf

Hafnium

73

Ta

Tantalum

74

W

Tungsten

75

Re

Rhenium

76

Os

Osmium

77

Ir

Iridium

78

Pt

Platinum

79

Au

Gold

80

Hg

Mercury

81

Tl

Thallium

82

Pb

Lead

83

Bi

Bismuth

84

Po

Polonium

85

At

Astatine

86

Rn

Radon

87

Fr

Francium

88

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Radium

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Ac

Actinium

90

Th

Thorium

91

Pa

Protactinium

92

U

Uranium

93

Np

Neptunium

94

Pu

Plutonium

95

Am

Americium

96

Cm

Curium

97

Bk

Berkelium

98

Cf

Californium

99

Es

Einsteinium

100

Fm

Fermium

101

Md

Mendelevium

102

No

Nobelium

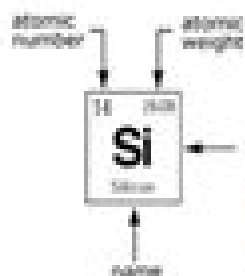
103

Lr

Lawrencium



LABS



black solid  
 blue liquid  
 red gas  
 white synthetically prepared  
 over water layer

- alkali metals
- alkaline earth metals
- transitional metals
- other metals
- nonmetals
- noble gases

|                     |                          |                       |                        |                       |                       |                        |                       |                         |                         |                      |                          |                       |                         |
|---------------------|--------------------------|-----------------------|------------------------|-----------------------|-----------------------|------------------------|-----------------------|-------------------------|-------------------------|----------------------|--------------------------|-----------------------|-------------------------|
| 58<br>Ce<br>Cerium  | 59<br>Pr<br>Praseodymium | 60<br>Nd<br>Neodymium | 61<br>Pm<br>Promethium | 62<br>Sm<br>Samarium  | 63<br>Eu<br>Europium  | 64<br>Gd<br>Gadolinium | 65<br>Tb<br>Terbium   | 66<br>Dy<br>Dysprosium  | 67<br>Ho<br>Holmium     | 68<br>Er<br>Erbium   | 69<br>Tm<br>Thulium      | 70<br>Yb<br>Ytterbium | 71<br>Lu<br>Lutetium    |
| 90<br>Th<br>Thorium | 91<br>Pa<br>Protactinium | 92<br>U<br>Uranium    | 93<br>Np<br>Neptunium  | 94<br>Pu<br>Plutonium | 95<br>Am<br>Americium | 96<br>Cm<br>Curium     | 97<br>Bk<br>Berkelium | 98<br>Cf<br>Californium | 99<br>Es<br>Einsteinium | 100<br>Fm<br>Fermium | 101<br>Md<br>Mendelevium | 102<br>No<br>Nobelium | 103<br>Lr<br>Lawrencium |

# Physical Science For Dummies

**J Elliott**



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laws of nature metaphysics of forces mathematics relation to nature and the concepts of space and time this book deepens our understanding of Hume's relation to natural philosophy It does so in addition by situating Hume's thought within the context of other major philosophers and scientists including Descartes Locke Boyle Kant Newton and Leibniz Demonstrating above all Hume's understanding of the fluid relationship between philosophy and science Hume's *Natural Philosophy and Philosophy of Physical Science* will provide new insights for historians and philosophers of science Chemical news and Journal of physical science ,1866 The Chemical News and Journal of Physical Science ,1904 I. *The Greek school philosophy, with reference to physical science.* II. *The physical sciences in ancient Greece.* III. *Greek astronomy.* IV. *Physical science in the middle ages.* V. *Formal astronomy after the stationary period.* VI. *Mechanics, including fluid mechanics.* VII. *Physical astronomy.* Additions to the 3d ed William Whewell,1858 *A Review of the Progress of Mathematical and Physical Science in More Recent Times, and Particularly between the Years 1775 and 1850* James D. Forbes,2025-08-27 Reprint of the original first published in 1858 The Antigonos publishing house specialises in the publication of reprints of historical books We make sure that these works are made available to the public in good condition in order to preserve their cultural heritage

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**Climate Change 2007 - The Physical Science Basis** Intergovernmental Panel on Climate Change, Intergovernmental Panel on Climate Change. Working Group I., 2007-09-10 The Climate Change 2007 volumes of the Fourth Assessment Report of the Intergovernmental Panel on Climate Change IPCC provide the most comprehensive and balanced assessment of climate change available This IPCC Working Group I report brings us completely up to date on the full range of scientific aspects of climate change Written by the world's leading experts the IPCC volumes will again prove to be invaluable for researchers students and policymakers and will form the standard reference works for policy decisions for government and industry worldwide

**Caesar, for beginners [selections from books 1, 4, 5 and 6] with notes by J. Currie** Gaius Julius Caesar, 1854

**Physics I For Dummies** Steven Holzner, 2022-04-12 An easy to follow guide to introductory physics from the Big Bang to relativity All science technology engineering and math majors in college and university require some familiarity with physics Other career paths like medicine are also only open to students who understand this fundamental science But don't worry if you find physics to be intimidating or confusing You just need the right guide In Physics I For Dummies you'll find a roadmap to physics success that walks you through every major topic in introductory physics including motion energy waves thermodynamics electromagnetism relativity and more You'll learn the basic principles and math formulas of physics through clear and straightforward examples and instruction and without unnecessary jargon or complicated theory In this book you'll also find Up to date examples and explanations appearing alongside the latest discoveries and research in physics discussed at a level appropriate for beginning students All the info found in an intro physics course arranged in an intuitive sequence that will give first year students a head start in their high school or college physics class The latest teaching techniques to ensure that you remember and retain what you read and practice in the book Physics I For Dummies is proof that physics can be fun accessible challenging and rewarding all at the same time Whether you're a high school or undergraduate student looking for a leg up on basic physics concepts or you're just interested in how our universe works this book will help you understand the thermodynamic electromagnetic relativistic and everything in between

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