

## Chapter 6 Chemical Bonds

**Section 6.2 Covalent Bonding****(pages 165–169)**

*This section discusses the formation of covalent bonds and the factors that determine whether a molecule is polar or nonpolar. It also discusses attractions between molecules.*

**Reading Strategy (page 165)**

**Relating Text and Visuals** As you read the section, look closely at Figure 9. Complete the table by describing each type of model shown. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

| Molecular Models   |             |
|--------------------|-------------|
| Model              | Description |
| Electron dot       |             |
| Structural formula |             |
| Space-filling      |             |
| Electron cloud     |             |

**Covalent Bonds (pages 165–167)**

- Describe a covalent bond. \_\_\_\_\_  
\_\_\_\_\_
- Circle the letters of molecular models that show orbitals of atoms overlapping when a covalent bond forms.
  - electron dot
  - structural formula
  - space-filling
  - electron cloud
- Describe a molecule. \_\_\_\_\_  
\_\_\_\_\_
- Is the following sentence true or false? In a covalent bond, the atoms are held together by the attractions between the shared electrons and the protons in each nucleus. \_\_\_\_\_
- Circle the correct answer. Nitrogen has five valence electrons. How many pairs of electrons must two nitrogen atoms share in order for each atom to have eight valence electrons?
  - zero
  - one
  - two
  - three

# Physical Science Chapter 6 Chemical Bonds Chapter Review

**LP Steffe**



## **Physical Science Chapter 6 Chemical Bonds Chapter Review:**

**Chemistry at the Frontier with Physics and Computer Science** Sergio Rampino, 2022-05-16 Chemistry at the Frontier with Physics and Computer Science Theory and Computation shows how chemical concepts relate to their physical counterparts and can be effectively explored via computational tools It provides a holistic overview of the intersection of these fields and offers practical examples on how to solve a chemical problem from a theoretical and computational perspective going from theory to models methods and implementation Sections cover both sides of the Born Oppenheimer approximation nuclear dynamics and electronic structure chemical reactions chemical bonding and cover theory to practice on three related physical problems wavepacket dynamics Hartree Fock equations and electron cloud redistribution Drawing on the interdisciplinary knowledge of its expert author this book provides a contemporary guide to theoretical and computational chemistry for all those working in chemical physics physical chemistry and related fields Combines a big picture overview of chemistry as it relates to physics and computer science including detailed guidance on tackling chemistry problems from both theoretical and computational perspectives Treats nuclear dynamics and electronic structure on the same footing in discussions of the Born Oppenheimer approximation Includes examples of scientific programming in modern Fortran for problems related to the modeling of chemical reaction dynamics and the analysis of chemical bonding

*Discover Science: Teacher's annotated edition*, 1991 Science content helps develop the skills needed to understand how science works learn new concepts solve problems and make decisions in today's technological society **Ebook:**

**Chemistry: The Molecular Nature of Matter and Change** Silberberg, 2015-01-16 Ebook Chemistry The Molecular Nature of Matter and Change **ATI TEAS Strategies, Practice & Review with 2 Practice Tests** Kaplan Nursing, 2017-01-03

Provides comprehensive exam review as well as test taking strategies and study techniques Addison-Wesley Introduction to Physical Science Michael B. Leyden, 1988 *Superatoms* Puru Jena, Qiang Sun, 2021-12-01 Explore the theory and

applications of superatomic clusters and cluster assembled materials *Superatoms Principles Synthesis and Applications* delivers an insightful and exciting exploration of an emerging subfield in cluster science superatomic clusters and cluster assembled materials The book presents discussions of the fundamentals of superatom chemistry and their application in catalysis energy materials science and biomedical sciences Readers will discover the foundational significance of superatoms in science and technology and learn how they can serve as the building blocks of tailored materials promising to usher in a new era in materials science The book covers topics as varied as the thermal and thermoelectric properties of cluster based materials and clusters for CO<sub>2</sub> activation and conversion before concluding with an incisive discussion of trends and directions likely to dominate the subject of superatoms in the coming years Readers will also benefit from the inclusion of A thorough introduction to the rational design of superatoms using electron counting rules Explorations of superhalogens endohedrally doped superatoms and assemblies and magnetic superatoms A practical discussion of atomically precise

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*Harcourt Science: Physical science, [grade] 5, Units E and F, teacher's ed*, 2000 **Misconceptions in Chemistry** Hans-Dieter Barke, Al Hazari, Sileshi Yitbarek, 2008-11-18 Over the last decades several researchers discovered that children pupils and even young adults develop their own understanding of how nature really works These pre concepts concerning combustion gases or conservation of mass are brought into lectures and teachers have to diagnose and to reflect on them for better instruction In addition there are school made misconceptions concerning equilibrium acid base or redox reactions which originate from inappropriate curriculum and instruction materials The primary goal of this monograph is to help teachers at universities colleges and schools to diagnose and cure the pre concepts In case of the school made misconceptions it will help to prevent them from the very beginning through reflective teaching The volume includes detailed descriptions of class room experiments and structural models to cure and to prevent these misconceptions **Glencoe Physical Science** McGraw Hill, 1999 2000 2005 State Textbook Adoption **Focus on Physical Science** Annette Miele Saturnelli, 1989 **Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science**, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction Chemistry 'O' Level Rex M. Heyworth, 2007

Springer Handbook of Surface Science Mario Rocca, Talat Rahman, Luca Vattuone, 2021-01-14 This handbook delivers an up to date comprehensive and authoritative coverage of the broad field of surface science encompassing a range of important materials such as metals semiconductors insulators ultrathin films and supported nanoobjects Over 100 experts from all branches of experiment and theory review in 39 chapters all major aspects of solid state surfaces from basic principles to applications including the latest ground breaking research results Beginning with the fundamental background of kinetics and thermodynamics at surfaces the handbook leads the reader through the basics of crystallographic structures and electronic properties to the advanced topics at the forefront of current research These include but are not limited to novel applications in nanoelectronics nanomechanical devices plasmonics carbon films catalysis and biology The handbook is an ideal reference guide and instructional aid for a wide range of physicists chemists materials scientists and engineers active

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