energy:	he element in each of the t	
a) lithium	or boron	c) cesium or lead
b) magnes	ium or strontium	d) sodium or argon
1. Arrange the	e following elements in ord	ler of increasing ionization energy:
a) Sr, Be, A	Mg	
b) Bi, Cs,	Ba	
c) Na, Al,	s	
2. Why is ther elements in		ween the 1st and 2nd ionization energies for the
Name:		Date: Class period:
	"lonic Size ar	Periodic Trends nd E.N. Trend" - WS #4
Pirections: Pla	ease answer each fill in the	e blank with the best answer. O
. A cation is	an ion with an overall	charge.
	The second secon	
. Cations are	made when an atom	<u>@</u>
		than the size of the neutral
. The size of		
. The size of atom of the	a positive ion is always e same element.	
. The size of atom of the . As the posi	a positive ion is always e same element. tive charge on an ion incre	than the size of the neutral
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. The size of atom of the . As the posi . An anion is . Anions are	a positive ion is always e same element. tive charge on an ion incre an ion with an overall made when at atom	ases, the size of the atom will charge.
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. The size of atom of the . As the position . An anion is . Anions are . The size of atom of the . As the negative . Which elemnicircle your a) Na or I	a positive ion is always e same element. tive charge on an ion incre an ion with an overall made when at atom a negative ion is always e same element. ative charge on an ion incre nent/ion has the biggest at answer. Na*1	than the size of the neutral ases, the size of the atom will charge than the size of the neutral eases, the size of the atom will comic/ionic size in each of the following pairs? Please d) Al or Al -3



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Table of Contents Periodic Trends Unit Exam And Answers

- 1. Understanding the eBook Periodic Trends Unit Exam And Answers
 - The Rise of Digital Reading Periodic Trends Unit Exam And Answers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Periodic Trends Unit Exam And Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Trends Unit Exam And Answers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Periodic Trends Unit Exam And Answers
 - Personalized Recommendations
 - Periodic Trends Unit Exam And Answers User Reviews and Ratings
 - Periodic Trends Unit Exam And Answers and Bestseller Lists
- 5. Accessing Periodic Trends Unit Exam And Answers Free and Paid eBooks
 - Periodic Trends Unit Exam And Answers Public Domain eBooks
 - Periodic Trends Unit Exam And Answers eBook Subscription Services
 - Periodic Trends Unit Exam And Answers Budget-Friendly Options

- 6. Navigating Periodic Trends Unit Exam And Answers eBook Formats
 - o ePub, PDF, MOBI, and More
 - Periodic Trends Unit Exam And Answers Compatibility with Devices
 - Periodic Trends Unit Exam And Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - o Adjustable Fonts and Text Sizes of Periodic Trends Unit Exam And Answers
 - Highlighting and Note-Taking Periodic Trends Unit Exam And Answers
 - Interactive Elements Periodic Trends Unit Exam And Answers
- 8. Staying Engaged with Periodic Trends Unit Exam And Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Trends Unit Exam And Answers
- 9. Balancing eBooks and Physical Books Periodic Trends Unit Exam And Answers
 - Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Periodic Trends Unit Exam And Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Trends Unit Exam And Answers
 - Setting Reading Goals Periodic Trends Unit Exam And Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Trends Unit Exam And Answers
 - Fact-Checking eBook Content of Periodic Trends Unit Exam And Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements

• Interactive and Gamified eBooks

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