

Periodic Table of the Elements

	+1	+2											+3	+4	+5	+6	+7	0
1	H 1.008	He 4.003											B 10.81	C 12.01	N 14.01	O 16.00	F 18.99	Ne 20.18
2	Li 6.94	Be 9.01											Al 26.98	Si 28.09	P 30.97	S 32.07	Cl 35.45	Ar 39.95
3	Na 22.99	Mg 24.31	Sc 44.96	Ti 47.88	V 50.94	Cr 52.00	Mn 54.94	Fe 55.85	Co 58.93	Ni 58.69	Cu 63.55	Zn 65.38	Ga 69.72	Ge 72.64	As 74.92	Se 78.96	Br 79.90	Kr 83.80
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9	Th 232.04	Pa 231.04	Np 237.04	Pu 244.06	Am 243.06	Cm 247.07	Bk 247.07	Cf 251.08	Es 252.08	Fm 257.10	Mn 258.10	No 259.10	Lr 262.10					

The only metal that is liquid at room temperature.

Period 3
element with
7 valence
electrons.

Semi-conductor
metalloid used
in electronics.

Only one valence electron, but not part of a family.

Extremely reactive non-metal.

Group that will form 2+ cations.

The only non-metal that is liquid at room temperature.

Completely
un-reactive family
with complete
valence shell.

Extremely reactive metal.

Group with 4 valence electrons.

Periodic Trends Review Answers

RM Cervero



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