

MYP4 Chemistry

Periodic Trends

Use the periodic table, charts, and your knowledge of periodic trends to answer the following questions.

Atomic Radius

1. Which atom in each pair has the larger atomic radius? Circle your answer.

a) Li or **K** b) **Ca** or Ni c) **Ga** or B d) O or **C** e) Cl or **Br**
f) Be or **Ba** g) **Si** or S h) Fe or **Au**

2. What is the periodic trend for atomic size from top to bottom in a group? From left to right in a period?

Atomic size (atomic radius) increases from top to bottom in a group and decreases from left to right in a period.

3. Why do atoms get smaller as you move left to right in a period?

Atoms get smaller as you move left to right in a period due to the increase in effective nuclear charge (while the number of occupied energy levels remain constant) leading to an increase in the force of attraction, which pulls the outer electrons closer to the nucleus and decreases the atomic radius.

4. Why do atoms get larger as you move down a group?

Atoms get larger as you move down a group due to the increase in occupied energy levels.

5. Arrange the following in order of increasing atomic radius atomic radius.

Mg, Na, P, Si and Ar? They are in the same period: Ar, P, Si, Mg, Na

Ionic Radius

6. Define an ion.

a charged atom or group of atoms

7. What is the difference between a cation and an anion?

Cations are positively charged ions. They are formed when an atom loses electrons.
Anions are negatively charged ions. They are formed when an atom gains electrons.

8. Explain the relationship between the relative size of an ion to its neutral atom and the charge on the ions.

Cations are smaller than their neutral atom because they have less occupied energy levels.
Anions are larger than their neutral atom because of the increase in electron-electron repulsion.

9. Which particle has the larger radius in each atom/ion pair? Circle your answer.

a) **Na** Na⁺ b) S, **S²⁻** c) I, I⁻ d) **Al** Al³⁺

Periodic Trends Practice Answer Key

L Reisser



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Problems and Problem Solving in Chemistry Education Georgios Tsaparlis, 2021-05-17

Problem solving is central to the teaching and learning of chemistry at secondary, tertiary and post tertiary levels of education, opening to students and professional chemists alike a whole new world for analysing data, looking for patterns and making deductions. As an important higher order thinking skill, problem solving also constitutes a major research field in science education. Relevant education research is an ongoing process with recent developments occurring not only in the area of quantitative computational problems but also in qualitative problem solving. The following situations are considered: some general others with a focus on specific areas of chemistry: quantitative problems, qualitative reasoning, metacognition and resource activation, deconstructing the problem solving process, an overview of the working memory hypothesis, reasoning with the electron pushing formalism, scaffolding organic synthesis skills, spectroscopy for structural characterization in organic chemistry, enzyme kinetics, problem solving in the academic chemistry laboratory, chemistry problem solving in context, team based active learning, technology for molecular representations, IR spectra simulation and computational quantum chemistry tools. The book concludes with methodological

and epistemological issues in problem solving research and other perspectives in problem solving in chemistry With a foreword by George Bodner The Boston Medical and Surgical Journal ,1927 **Chemistry in the Community (ChemCom)** American Chemical Society,2011-06-17 Touted as the most successful NSF funded project published Chemistry in the Community ChemCom by the American Chemical Society ACS offers a meaningful and memorable chemistry program for all levels of high school students ChemCom covers traditional chemistry topics within the context of societal issues and real world scenarios Centered on decision making activities where students are responsible for generating data in an investigating analyzing that data and then applying their chemistry knowledge to solve the presented problem The text is intensively laboratory based with all 39 of the investigations integrated within the text not separate from the reading With the ChemCom program students learn more organic and biochemistry more environmental and industrial chemistry and more on the particulate nature of matter than other textbooks all within the relevance of solving problems that arise in everyday life Meticulously updated to meet the needs of today s teachers and students the new sixth edition of ChemCom adheres to the new science framework as well as the forthcoming next generation of science standards Incorporating advances in learning and cognitive sciences ChemCom s wide ranging coverage builds upon the concepts and principles found in the National Science Education Standards Correlations are available showing how closely aligned ChemCom is to these and other state standards *Oswal - Gurukul Chemistry Most Likely Question Bank : ICSE Class 10 For 2023 Exam* Oswal - Gurukul,2022-05-14 PTCB Exam Study Guide 2025-2026 Jennie Victoria Davenport ,Judith Bess Perkins, PTCB Exam Study Guide 2025 2026 600 Practice Questions with Detailed Answer Explanations for the Pharmacy Technician Certification Board Test Discover the ultimate study guide for the Pharmacy Technician Certification Board PTCB Exam This comprehensive book is designed to help you prepare for and pass the PTCB Exam in 2025 2026 launching your career as a certified pharmacy technician Packed with 600 practice questions that mirror the actual PTCB Exam content this study guide provides you with the opportunity to test your knowledge and familiarize yourself with the exam format Each practice question comes with a detailed answer explanation helping you understand the reasoning behind the correct answer and learn from your mistakes The book covers all the essential topics you need to know for the PTCB Exam including Pharmacology Pharmacy law and regulations Sterile and non sterile compounding Medication safety Pharmacy quality assurance Medication order entry and fill process Pharmacy inventory management Pharmacy billing and reimbursement In addition to the practice questions and content review this study guide offers proven test taking strategies and tips to help you approach the exam with confidence You ll also find valuable insights into the roles and responsibilities of pharmacy technicians in various practice settings as well as practical advice for launching and advancing your career This PTCB Exam study guide is an indispensable resource for anyone seeking to pass the Pharmacy Technician Certification Board Exam in 2025 2026 Aspiring pharmacy technicians and experienced professionals alike will benefit from the comprehensive content

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