				Score	/100
Mil in receiver	REV	Period	Thursday :		

Fill in the blank (2 points each blank)

- Elements that have the properties of both metal and non-metals are metalloids.
- 2. What do elements in the same group have the same number of? Valence electrons
- 3. Which element would have the highest electron affinity? Fluorine
- 4. What element is at period 3, group 8A? Argon
- 5. How many valence electrons does Bromine have? Z
- 6. The atomic number tells you how many protons (and electrons if neutral) there are in an atom.
- 7. Two Characteristics of Non-metals are not good conductors of heat or electricity, not malleable, etc.
- 8. Two characteristics of Metals are malleable, ductile, good conductors of heat and electricity
- Which element would have the smallest atomic radius? Helium
- Which element would have the highest ionization energy? Helium

11. For the questions below and based on the information provided, till in the missing information. (1 point each)

Name	Symbol	# of protons	# of electrons	# of neutrons	Atomic Number	Moss Number
Potassium	K	19	15	22	15	-41
Phosphorous	p	1.5	15	18	15	33
Beryllium	Be	4	-4	4	4	8
Sulfur	S	16	16	20	16	3.6

- 12. Draw Bohr Models for (3 points each)
 - a. Na23

b. N15

13. What is an isotope? (1 point)

An atom with a different number of neutrons

14. How can you tell the identity of any atom? (1 point)

Number of PROTONS

15. Alien Periodic Chart -- Directions: Read the clue and place the alien symbol in the corresponding box!

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Table of Contents Periodic Table Test A Answer Key

- 1. Understanding the eBook Periodic Table Test A Answer Key
 - The Rise of Digital Reading Periodic Table Test A Answer Key
 - o Advantages of eBooks Over Traditional Books
- 2. Identifying Periodic Table Test A Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Test A Answer Key
 - User-Friendly Interface

- 4. Exploring eBook Recommendations from Periodic Table Test A Answer Key
 - Personalized Recommendations
 - Periodic Table Test A Answer Key User Reviews and Ratings
 - Periodic Table Test A Answer Key and Bestseller Lists
- 5. Accessing Periodic Table Test A Answer Key Free and Paid eBooks
 - Periodic Table Test A Answer Key Public Domain eBooks
 - Periodic Table Test A Answer Key eBook Subscription Services
 - Periodic Table Test A Answer Key Budget-Friendly Options
- 6. Navigating Periodic Table Test A Answer Key eBook Formats
 - o ePub, PDF, MOBI, and More
 - Periodic Table Test A Answer Key Compatibility with Devices
 - Periodic Table Test A Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table Test A Answer Key
 - Highlighting and Note-Taking Periodic Table Test A Answer Key
 - Interactive Elements Periodic Table Test A Answer Key
- 8. Staying Engaged with Periodic Table Test A Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Test A Answer Key
- 9. Balancing eBooks and Physical Books Periodic Table Test A Answer Key
 - Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Periodic Table Test A Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Table Test A Answer Key
 - Setting Reading Goals Periodic Table Test A Answer Key
 - Carving Out Dedicated Reading Time

- 12. Sourcing Reliable Information of Periodic Table Test A Answer Key
 - Fact-Checking eBook Content of Periodic Table Test A Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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