

# PERIODIC TABLE (ADVANCED)

## Molecular Structure of Hydrogen



### Basic Information

Name: Hydrogen  
Symbol: H  
Atomic Number: 1  
Atomic Mass: 1.00794 amu  
Melting Point:  $-259.14^{\circ}\text{C}$  ( $-314.009165^{\circ}\text{K}$ ,  $-434.41203^{\circ}\text{F}$ )  
Boiling Point:  $-252.87^{\circ}\text{C}$  ( $-29.280005^{\circ}\text{K}$ ,  $-423.166^{\circ}\text{F}$ )  
Number of Protons/Electrons: 1  
Number of Neutrons: 0  
Classification: Non-metal  
Crystal Structure: Hexagonal  
Density @ 293 K: 0.08989 g/cm<sup>3</sup>  
Color: colorless

### Atomic Structure

Number of Energy Levels: 1  
First Energy Level: 1

### Isotopes

Isotope	Half-life
H-1	Stable
H-2	Stable
H-3	12.3 years

### Facts

Name Origin: from words water (hydro) and generate (genesis)  
Uses: balloons, metal refining  
Obtained From: mines, oil, gas wells

## Molecular Structure of Carbon



### Name: Carbon

Symbol: C  
Atomic Number: 6  
Atomic Mass: 12.0107 amu  
Melting Point:  $3500.0^{\circ}\text{C}$  ( $3773.15^{\circ}\text{K}$ ,  $6332.0^{\circ}\text{F}$ )  
Boiling Point:  $4827.0^{\circ}\text{C}$  ( $5100.15^{\circ}\text{K}$ ,  $8700.6^{\circ}\text{F}$ )  
Number of Protons/Electrons: 6  
Number of Neutrons: 6  
Classification: Non-metal  
Crystal Structure: Hexagonal  
Density @ 293 K: 2.62 g/cm<sup>3</sup>  
Color: May be black

### Atomic Structure

Number of Energy Levels: 2  
First Energy Level: 2      Second Energy Level: 4

### Isotopes

Isotope	Half-life
C-11	20.3 minutes
C-12	Stable
C-13	Stable
C-14	5730.0 years
C-15	2.5 seconds

### Facts

Uses: steel, fibers  
Obtained From: burning with insufficient oxygen

# Periodic Table Advanced Speedy Study Guide

**Clemens Wendtner**



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