

Colligative Properties Worksheet

FYI: K_f of water is 1.86°C/m ; K_b of water is 0.52°C/m

- 1) Explain why the addition of a solute decreases the melting point of a liquid.
- 2) What is the melting point of a solution in which 3.5 grams of sodium chloride is added to 230 mL of water?
- 3) What is the boiling point of a solution in which 22.5 grams of sucrose ($\text{C}_{12}\text{H}_{22}\text{O}_{11}$) is dissolved in 550 mL of water?
- 4) Rank the following solutions from lowest to highest boiling point, a 2.5 m solution of sodium chloride, a 3.5 m solution of magnesium chloride, or a 4.5 m solution of sulfur dioxide?
- 5) If an aqueous solution has a boiling point of 102.5° , what's the effective molality of the solute?

Practice Test On Colligative Properties

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A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the American BookWorks Corporation text.

Practice Test On Colligative Properties:

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With this objective in mind we are presenting before you this book containing unit tests Some features of the books are The complete syllabus is divided into logical units and there is a self assessment tests for each unit Tests are prepared by subject experts who have decade of experience to prepare students for competitive exams Tests are as per the latest pattern of the examination Detailed explanatory solution of each test paper is also given Student is advised to attempt these Tests once they complete the preparation revision of unit They should attempt these Test in exam like environment in a specified time Student is advised to properly analyze the solutions and think of alternative methods and linkage to the solutions of identical problems also We firmly believe that the book in this form will definitely help a genuine hardworking student We have put our best efforts to make this book error free still there may be some errors We would appreciate if the same is brought to our notice We wish to utilize the opportunity to place on record our special thanks to all faculty members and editorial team for their efforts to make this book

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faculty members and editorial team for their efforts to make this book

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