Periodic Trends Practice

Alonis Redii

The distance from the necleoses of two bonded atoms



Electronogativity

How much an atom. "wants" electrons on the atom's fundancy to attract electrons.



Energy

The amount of energy required to remove electrons from an atom.



Electron Affinity

The amount of energy released when a neutral gaseous atom gains an electron.



Driving Question

How can the periodic table be used to predict atomic characteristics such as atomic radius, electronogetivity, insignation energy, and electron affinity.

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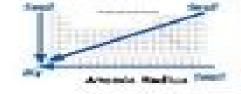
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Shielding?

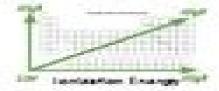
What is shielding? What effect does it have on the periodic trends defined above?

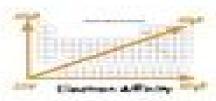
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Add arress to each of the following periods tables to indicate the direction of the trend as it increases or decreases from left to right and top to bettern. If the Dirichlet solds will be sected to be provided to be you can easily remember their









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