

Name _____

Physical Science Chemical Bonding Study Guide

A. Covalent & Ionic Compounds/Bonds

1. Which ion accepts electrons, and which from it? (anion/cation)
which cation it gains and electron it loses from it? (anion/cation)
2. Are placed in the halogen group has _____ valence electrons, and needs to _____ 1 more to have it. This is because it has _____ valence.
3. Are _____ bond to the adjacent elements oppositely charged ions.
PCl₃ is covalent bond is _____
4. Is a covalent bond, shared _____, electron, shared by all
ionic bond, electron _____, electron.
5. (Electron group) Which side from left shared completely is a covalent bond is to _____, and which group is shared equally to the _____
How can predict if a compound is ionic, polar covalent, or nonpolar covalent by
looking at the _____
6. Are ionic ions have up to _____ valence electrons. This usually indicates an ionic
compound. Ionic ions all have 8 valence electrons in shell.
7. Shared pair gets from _____, or shared _____ by both atoms.

B. Formulas for Ionic compounds - Put in the right order

Write correct electron configuration in	Magnesium	Fluorine	Phosphorus	Lithium

3. Write in the chemical formula for each of the following ions:
compounds of

Draw your own ionic bond structures, and use arrows to show the transfer of electrons. Use your periodic table to find their oxidation numbers. They will combine so that their oxidation numbers cancel out to zero. Use the criss cross method if their oxidation numbers do not cancel.

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