

## 11

## CHEMICAL REACTIONS

## Practice Problems

*In your notebook, solve the following problems. Use the 3-step problem-solving approach you learned in Chapter 1.*

## SECTION 11.1 DESCRIBING CHEMICAL REACTIONS

1. Write the skeleton equation for the reaction between hydrogen and oxygen that produces water.
2. Write the skeleton equation for the reaction that produces iron(II) sulfide from iron and sulfur.
3. Write the skeleton equation representing the heating of magnesium carbonate to produce solid magnesium oxide and carbon dioxide gas.
4. Write a balanced equation for the production of HCl gas from its elements.
5. Write a sentence that completely describes the chemical reaction represented by this balanced equation.



6. Write the word equation for the following equation. Write a sentence fully describing the reaction. Is the equation correctly balanced? Explain.



7. Write a balanced equation representing the formation of aqueous sulfuric acid from water and sulfur trioxide gas.
8. Write a balanced equation from this word equation.  
aqueous silver nitrate + copper metal  $\rightarrow$  silver metal + aqueous copper nitrate
9. Write a balanced equation for the following word equation.



## SECTION 11.2 TYPES OF CHEMICAL REACTIONS

1. Write a balanced equation representing the reaction of magnesium with oxygen gas to produce magnesium oxide.
2. Write the balanced equation for the reaction that occurs between aluminum and fluorine.
3. Write the balanced equation for the production of oxygen gas and potassium chloride from the decomposition of potassium chlorate.
4. Write the balanced equation for the reaction between hydrochloric acid and calcium metal. The products are hydrogen gas and calcium chloride.
5. Write the balanced equation for the combustion of propane ( $\text{C}_3\text{H}_8$ ) to produce carbon dioxide and water vapor.
6. Write the balanced equation for the reaction between iron(III) chloride and sodium hydroxide. The products are iron(III) hydroxide and sodium chloride.

# Practice Problems Describing Chemical Reactions

**Laurene V. Fausett**



## **Practice Problems Describing Chemical Reactions:**

**The Practice of Chemistry Study Guide & Solutions Manual** Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises **ATI TEAS 7 Crash Course with Online Practice Test, 4th Edition** John Allen, 2024-01-11

Everything you need for today's ATI TEAS Version 7 in a concise time saving format Provided by publisher **The Practice of Chemistry** Donald J. Wink, Sharon Fetzer-Gislason, Sheila McNicholas, 2003-03 Students can't do chemistry if they can't do the math The Practice of Chemistry First Edition is the only preparatory chemistry text to offer students targeted consistent mathematical support to make sure they understand how to use math especially algebra in chemical problem solving The book's unique focus on actual chemical practice extensive study tools and integrated media makes The Practice of Chemistry the most effective way to prepare students for the standard general chemistry course and bright futures as science majors

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**Linear Partial Differential Equations and Fourier Theory** Marcus Pivato, 2010-01-07 Do you want a rigorous book that remembers where PDEs come from and what they look like This highly visual introduction to linear PDEs and initial boundary value problems connects the math to physical reality all the time providing a rigorous mathematical foundation for all solution methods Readers are gradually introduced to abstraction the most powerful tool for solving problems rather than simply drilled in the practice of imitating solutions to given examples The book is therefore ideal for students in mathematics and physics who require a more theoretical treatment than given in most introductory texts Also designed with lecturers in mind the fully modular presentation is easily adapted to a course of one hour lectures and a suggested 12 week syllabus is included to aid planning Downloadable files for the hundreds of figures hundreds of challenging exercises and practice problems that appear in the book are available online as are solutions *Multiple Representations in Chemical Education*

John K. Gilbert, David Treagust, 2009-02-28 Chemistry seeks to provide qualitative and quantitative explanations for the observed behaviour of elements and their compounds Doing so involves making use of three types of representation the macro the empirical properties of substances the sub-micro the natures of the entities giving rise to those properties and the symbolic the number of entities involved in any changes that take place Although understanding this triplet relationship is a key aspect of chemical education there is considerable evidence that students find great difficulty in achieving mastery of the ideas involved In bringing together the work of leading chemistry educators who are researching the triplet relationship at the secondary and university levels the book discusses the learning involved the problems that students encounter and successful approaches to teaching Based on the reported research the editors argue for a coherent model for understanding the triplet relationship in chemical education *Mathematical Models of Chemical Reactions* Péter Érdi, János Tóth, 1989

**LSENS, a General Chemical Kinetics and Sensitivity Analysis Code for Homogeneous Gas-phase Reactions. 2:**

**Code Description and Usage** Krishnan Radhakrishnan,1994      **CliffsNotes Chemistry Practice Pack** Charles Henrickson,2010-02-08 About the Contents Pretest Helps you pinpoint where you need the most help Topic Area Reviews Measurement and Units of Measurement Matter Elements Compounds and Mixtures Atoms I The Basics Formulas and Names of Ionic Compounds Acids and Bases The Mole Elements and Compounds Percent Composition and Empirical and Molecular Formulas Chemical Reactions and Chemical Equations Calculations Using Balanced Equations Atoms II Atomic Structure and Periodic Properties Chemical Bonding The Formation of Compounds Gases and the Gas Laws The Forces between Molecules Solids and Liquids Solutions and Solution Composition Acids Bases and Neutralization Glossary Customized Full Length Exam Covers all subject areas Pretest that pinpoints what you need to study most Clear concise reviews of every topic Targeted example problems in every chapter with solutions and explanations Customized full length exam that adapts to your skill level      **NASA Tech Briefs** ,1995      **Interface Problems and Methods in Biological and Physical Flows** Boo Cheong Khoo,Zhilin Li,Ping Lin,2009 This volume showcases lecture notes collected from tutorials presented at the Workshop on Moving Interface Problems and Applications in Fluid Dynamics that was held between January 8 and March 31 2007 at the Institute for Mathematical Sciences National University of Singapore As part of the program these tutorials were conducted by specialists within their respective areas such as Robert Dillon Zhilin Li John Lowengrub Frank Lu and Gretar Tryggvason The topics in the program encompass modeling and simulations of biological flow coupled to deformable tissue elastic structure shock wave and bubble dynamics and various applications like biological treatments with experimental verification multi medium flow or multiphase flow and various applications including cavitation supercavitation detonation problems Newtonian and non Newtonian fluid and many other areas This volume stand to benefit graduate students and researchers keen in the field of interfacial flows for application to physical and biological systems Even beginners will find this volume a very useful starting point with many relevant references applicable      **General Organic and Biological Chemistry** Kenneth W. Raymond,2009-12-14 This general organic and biochemistry text has been written for students preparing for careers in health related fields such as nursing dental hygiene nutrition medical technology and occupational therapy It is also suited for students majoring in other fields where it is important to have an understanding of the basics of chemistry Students need have no previous background in chemistry but should possess basic math skills The text features numerous helpful problems and learning features      Scaling Analysis in Modeling Transport and Reaction Processes William B. Krantz,2007-06-30 This book is unique as the first effort to expound on the subject of systematic scaling analysis Not written for a specific discipline the book targets any reader interested in transport phenomena and reaction processes The book is logically divided into chapters on the use of systematic scaling analysis in fluid dynamics heat transfer mass transfer and reaction processes An integrating chapter is included that considers more complex problems involving combined transport phenomena Each chapter includes several problems that are explained in

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mathematical background in sufficient detail for this highly current and important topic but without unnecessary complications. They focus on approaches with industrial relevance covering real life applications taken from concrete projects that range from tribology modeling to performance optimization of integrated circuits. Following an introduction to the fundamentals, the book describes the most relevant approaches covering such classical simulation methods as simple and reactive force field methods as well as highly accurate quantum mechanical methods ranging from density functional theory to Hartree Fock and beyond. A review of the increasingly important multiscale approaches rounds off this section. The last section demonstrates and illustrates the capabilities of the methods previously described using recent real life examples of industrial applications. As a result, readers gain a heightened user awareness since the authors clearly state the conditions of applicability for the respective modeling methods so as to avoid fatal mistakes.

Numerical Methods Using MathCAD  
Laurene V. Fausett, 2002. This book presents the fundamental numerical techniques used in engineering, applied mathematics, computer science, and the physical and life sciences in a way that is both interesting and understandable. Using a wide range of examples and problems, this book focuses on the use of MathCAD functions and worksheets to illustrate the methods used when discussing the following concepts: solving linear and nonlinear equations, numerical linear algebra, numerical methods for data interpolation and approximation, numerical differentiation and integration, and numerical techniques for solving differential equations. For professionals in the fields of engineering, mathematics, computer science, and physical or life sciences who want to learn MathCAD functions for all major numerical methods.

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