

Name: _____

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Chapter 5 Quiz Biology

1. The ability to do work
 2. Stored energy that can be used for motion
 3. Energy of motion
 4. Energy conversion to heat, Heat energy
 5. The study of energy or heat
 6. Can never be created nor destroyed, amount of energy, in the universe remains constant
 7. Amount of energy transforming from potential to heat energy, or entropy in universe is increasing
 8. The starting molecules of a chemical reactions
 9. Molecules at the end of a chemical reaction
 10. Reactions have products w/ more energy than the reactants (uses & stores energy) spontaneous
 11. An initial input of energy by destabilizing existing chemical bonds
 12. reactions become more spontaneous if their activation is lowered
 13. Catalysts used by cells to perform particular reactions
 14. Enzymes lower activation energy by binding specific molecules and stressing the bonds to make the reaction more likely to proceed
 15. includes all the chemical and physical process continuously occurring in living cells
 16. Is the process by which food is changed into living tissue
 17. Is the process by which food is converted into energy and wastes of a simpler chemical composition
 18. Series of reactions of the product of one reaction is the substrate of the next
 19. What do free "inorganic" phosphates do?
 20. When an atom or molecule loses an electron
 21. When an atom or molecule gain an electron
 22. Have "the power" causes something else to be reduced
 23. Where is energy "Stored" in ATP molecule?
 24. electrons pass from atoms or molecules to one another as part of many energy reactions
 25. Provide energy for the initiation of endergonic reactions
- A. Activation energy (enzymes)
 - B. Enzymes
 - C. Catabolism
 - D. Anabolism
 - E. Potential energy
 - F. 1st law of thermodynamics
 - G. Reducing power
 - H. Reactants and Substrates
 - I. Oxidation
 - J. Products
 - K. B/w phosphate bonds
 - L. Thermodynamics
 - M. Endergonic
 - N. Coupled reactions
 - O. Biochemical Pathways
 - P. Redox Reduction
 - Q. Catalysis
 - R. Release energy
 - S. Activation energy
 - T. Energy
 - U. How energy is measured
 - V. Kinetic energy
 - W. Reduction
 - X. 2nd law of thermodynamics
 - Y. Metabolism

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Michael Hoelscher



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