

Chapter 7: Chemical Reaction Guided Notes

I. Chemical Changes in Matter

A. Chemical Reaction

1. A change in which one or more substances are converted to _____ substances.

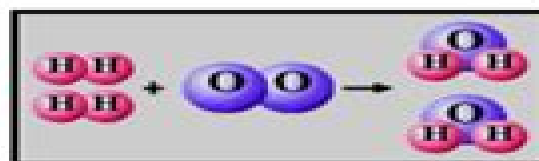


B. Law of Conservation of Matter

1. In a chemical reaction, matter is not _____ or _____.
2. Atoms can only be _____.
3. Discovered by Lavoisier.

_____ Hydrogen

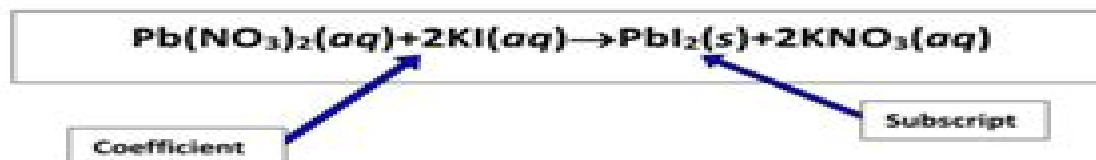
_____ Oxygen



_____ Hydrogen

_____ Oxygen

C. Chemical Equations



1. _____ - # of units of each substance.
2. Individual atom = **atom**
 $2\text{Mg} \Rightarrow 2$ _____ of magnesium
3. Covalent substance = **molecule**
 $3\text{CO}_2 \Rightarrow 3$ _____ of carbon dioxide
4. Ionic substance = **unit**
 $4\text{MgO} \Rightarrow 4$ _____ of magnesium oxide

SYMBOL	MEANING
$\longrightarrow$	produces, forms
+	plus, and
(s)	solid
(l)	liquid
(g)	gas
(aq)	aqueous (solid dissolved in water)
Δ	the reactants are heated

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IM Harris



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Table of Contents **Notetaking Guide Answer Keychemical Reactions**

1. Understanding the eBook **Notetaking Guide Answer Keychemical Reactions**
 - The Rise of Digital Reading **Notetaking Guide Answer Keychemical Reactions**
 - Advantages of eBooks Over Traditional Books
2. Identifying **Notetaking Guide Answer Keychemical Reactions**
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an **Notetaking Guide Answer Keychemical Reactions**
 - User-Friendly Interface
4. Exploring eBook Recommendations from **Notetaking Guide Answer Keychemical Reactions**
 - Personalized Recommendations
 - **Notetaking Guide Answer Keychemical Reactions** User Reviews and Ratings
 - **Notetaking Guide Answer Keychemical Reactions** and Bestseller Lists

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 - Highlighting and Note-Taking Notetaking Guide Answer Keychemical Reactions
 - Interactive Elements Notetaking Guide Answer Keychemical Reactions
8. Staying Engaged with Notetaking Guide Answer Keychemical Reactions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Notetaking Guide Answer Keychemical Reactions
9. Balancing eBooks and Physical Books Notetaking Guide Answer Keychemical Reactions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Notetaking Guide Answer Keychemical Reactions
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Notetaking Guide Answer Keychemical Reactions
 - Setting Reading Goals Notetaking Guide Answer Keychemical Reactions
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Notetaking Guide Answer Keychemical Reactions
 - Fact-Checking eBook Content of Notetaking Guide Answer Keychemical Reactions
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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