

Released Question from Chemistry EOC

- 5 Element X has two known naturally occurring isotopes. The mass and relative abundance of each isotope are shown below.

Relative Abundance	Mass (amu)
50.57%	78.92
49.43%	80.92

What is the average atomic mass of Element X to the nearest hundredth of an atomic mass unit?

Record your answer and fill in the bubbles on your answer document.

Step 1: Convert % to Decimals.

Step 2: Multiple Relative Abundance by Mass.

Step 3: Add the numbers for final answer.

Step 4: Round correctly to the nearest hundredth and grid answer.

Released Chemistry Eoc

G Thomas



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