

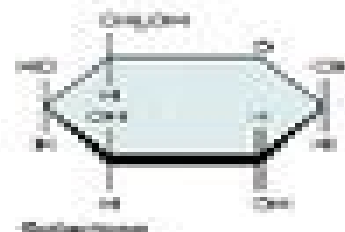
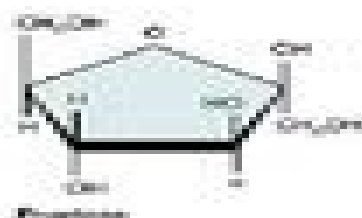
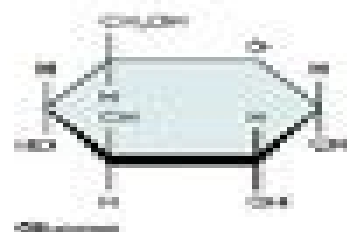
Biological Macromolecules (OpenStax Biology, Chapter 3) - Reinforcement (KIR)

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1. What are the four macromolecules important to life? carbohydrate, lipids, proteins, nucleic acids
2. Monomers combine with each other to create polymers
3. The creation of these large molecules releases water, the reaction is called: dehydration synthesis
4. What type of bond is created during this reaction? covalent
5. Polymers can be broken down into individual monomers. This process is called hydrolysis
6. What can speed up these reactions? enzymes

3.2 Carbohydrates

7. What is the ratio of carbon, hydrogen and oxygen in carbohydrates? 1:2:1
8. What is an example of a pentose monosaccharide? ribose Hexose? glucose
9. What is the chemical formula for glucose? $C_6H_{12}O_6$
10. In plants, excess glucose is stored as starch
11. Examine each of the monosaccharides below identified as hexoses.



- a) What is the molecular formula for fructose? $C_6H_{12}O_6$
 - b) Molecules with the same molecular formula but differ structurally are called isomers
 - c) Why are these molecules called hexoses? 6 carbons
12. Two monosaccharides can combine to create disaccharides
 13. A covalent bond that forms between two monosaccharides is called a glycosidic bond.
 14. The most common disaccharide is sucrose (table sugar)
 15. Identify the monosaccharide subunits in each molecule.
Sucrose = fructose + glucose Lactose = glucose + galactose

Reinforcement Biological Molecules Answer Key

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