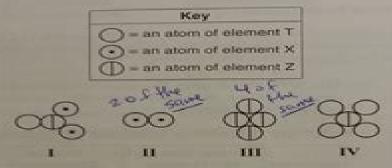
39 Given four particle models:



Which two models can be classified as elements?

(1) Land III

(3) II and III

(2) Land IV

- (4) II and IV
- 40 After being thoroughly stirred at 10.°C, which mixture is heterogenous?



- 25.0 g of KNO, and 100. g of HO Supersat.
- (3) 25.0 g of NaCl and 100. g of H₂O VOSAT.
- (4) 25.0 g of NaNO, and 100. g of H-O unsAT-
- 41 Which two compounds are electrolytes?

(1) KOH and CH-COOH

(2) KOH and CoH12

- (3) CH₃OH and CH₃COOH
- (4) CH₂OH and C₂H₁₂
- 42 Which statement explains why a CO₂ molecule is mongrolar?
 - Carbon and oxygen are both nonmetals.
 - (2) Carbon and oxygen have different electronegativities.
 - (3) The molecule has a symmetrical distribution of claurge.
 - (4) The molecule has an asymmetrical distribution of charge.
- 43 Which temperature change indicates an increase in the average kinetic energy of the molecules in a sample?
 - (B) 15 C to 298 K T
- (3) 305 K to 0°C
- 0 37°C to 273 K A
- (4) 355 K to 25°C
- PARTICIPATION FROM THE

44 Given the particle diagram:

Key = an atom



Which substance at STP can be represented by this particle diagram? Soli d

(1) No (0)

(3) Mg(5)

(2) H₂ (0)

(4) Kr (a)

45 Which type of equilibrium exists in a sealed flask containing Br₂(£) and Br₂(g) at 298 K and 1.0 atm?

- (1) static phase equilibrium
- (2) static solution equilibrium
- (3) dynamic phase equilibrium
- (4) dynamic solution equilibrium

46 What are the products when potassium hydroxide reacts with hydrochloric acid? Mark

- KH(s), Cl⁺(aq), and OH⁻(aq)
- (2) K(s), Cl₂(g), and H₂O(ℓ)
- (3) KCl(aq) and H₂O(ε)
- (4) KOH(aq) and Cl₂(g)

47 In a titration, 20.0 milliliters of a 0.150 M NaOH(aq) solution exactly neutralizes 24.0 milliliters of an HCl(aq) solution. What is the concentration of the HCl(aq) solution?

(1) 0.125 M

(3) 0.250 M

(2) 0.180 M

(4) 0.360 M

48 What fraction of a Sr-90 sample remains unchanged after 87,3 years?







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- 1. Understanding the eBook Regents Chemistry Prentice Hall Topic 4 Answers
 - The Rise of Digital Reading Regents Chemistry Prentice Hall Topic 4 Answers
 - Advantages of eBooks Over Traditional Books
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 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Regents Chemistry Prentice Hall Topic 4 Answers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Regents Chemistry Prentice Hall Topic 4 Answers
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 - Regents Chemistry Prentice Hall Topic 4 Answers and Bestseller Lists
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 - Interactive Elements Regents Chemistry Prentice Hall Topic 4 Answers
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 - Participating in Virtual Book Clubs
 - \circ Following Authors and Publishers Regents Chemistry Prentice Hall Topic 4 Answers
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 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Regents Chemistry Prentice Hall Topic 4 Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Regents Chemistry Prentice Hall Topic 4 Answers
 - Setting Reading Goals Regents Chemistry Prentice Hall Topic 4 Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Regents Chemistry Prentice Hall Topic 4 Answers
 - Fact-Checking eBook Content of Regents Chemistry Prentice Hall Topic 4 Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
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