#### Regents Chemistry Unit 8 Thermochemistry Practice Test

- At which temperature do both the solid and the liquid phase of water exist?
  - (1) 273 °C (2) 100 °C.

- (3) 0°C (4) 373 °C
- 2. What is the freezing point of sodium?
  - (I) 371°C (2) 644 °C

of iron?

- (3) -98 °C (4) 98 °C
- 3. A 15.75-g piece of iron absorbs 1086.75 joules of heat energy, and its temperature changes from 25°C to 175°C. What is the specific heat capacity

  - (1) 1.0 J/g\*\*C (3) 0.46 J/g\*\*C
  - (2) 4.18 1/g+0C (4) 2.76 1/g+0C
- 4. Which statement concerning propanone and water at 50°C is true?
  - (1) Propanone has a lower vapor pressure and weaker intermolecular forces than water
  - (2) Propanine has a lower vapor pressure and stronger intermolecular forces than water.
  - (3) Propanone has a higher vapor pressure and weaker intermolecular forces than water.
  - (4) Propanone has a higher vapor pressure and stronger intermolecular forces than water.
- Which phase change is an exothermic process?
  - (I) CO<sub>2</sub> (s) → CO<sub>2</sub> (g)
- (2) Cia (s)--- Cu(A
- (3) NH<sub>3</sub>(I) → NH<sub>3</sub> (g)
- (4) Hg (A→ Hg (s)
- 6. What is the vapor pressure of water at 105°C?
  - (1) 100 kPa.

(3) 101.3 kPa

(2) 110 kPa

- (4) 120 kPa
- 7. As a substance changes from a liquid to a gas, the average distance between molecules
  - (II) decreases.
  - (2) increases
  - (3) remains the same

8. The graph below represents the uniform heating of a substance, starting with the substance as a solid below its melting point.



Which line segment represents an increase in average kinetic energy and so change in potential energy?

(I) BC

(3) ED

(2) AB

- (4) IDE:
- How many joules are required to melt 100, grams of water?
  - (1) 226,000 kJ
- (2)  $22,600 \pm$
- (3) 33,400 kJ
- (4) 33,400 [
- 10. What occurs when a 35-gram aluminum cube at 100,°C is place in 90, grams of water at 25°C in an insulated cup?
  - (1) Heat is transferred from the aluminum to the water, and the temperature of the water decreases.
  - (2) Heat is transferred from the aluminum to the water, and the temperare of the water increases.
  - (3) Heat is transferred from the water to the aluminum, and the temperature of the water decreases.
  - (4) Heat is transferred from the water to the aluminum, and the temperature of the water increases.
- 11. What is the boiling point of propanone if the pressure on its surface is 48 kilopascals?
  - (I) 25°C

(3) 35°C

(2) 30.°C

(4) 40.°C

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University of California (System).
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Report of the Board of Regents California. University. Regents, 1917

Biennial Report of the President of the University on Behalf of the Regents ... University of California (System). Regents, 1891

University on Behalf of the Regents California. University. Regents, 1882

Annual Report of the President of the University of California University of California University of California,

Berkeley, 1891

Biennial Report of the President of the University on Behalf of the Board of Regents University of California (System), 1884

Catalogs of Courses University of California, Berkeley, 1994 Includes general and summer catalogs issued between 1878 1879 and 1995 1997

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American Men and Women of Science ,1977 Annual Report of the President of the University on Behalf of the Regents to His Excellency the Governor of the State of California University of California (1868-1952).

President,1917 Regents Chemistry-Physical Setting Power Pack Revised Edition Albert S. Tarendash,2021-01-05

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