Unit 5, Activity 5

Balancing Redox Reactions using half-reaction method worksheet

Balance the following equations using the half-reaction method. Identify the reactant being oxidized and reduced.

a.
$$Cu^{2+} + I^{1-} \rightarrow Cu + I_3^{1-}$$
 (neutral)

b.
$$MnO_4^{1-} + Ag \rightarrow Mn^{2+} + Ag^{1+}$$
 (acidic)

c.
$$Hg + NO_3^{1-} + Cl^{1-} \rightarrow HgCl_4^{2-} + NO_2$$
 (acidic)

d.
$$AsH_3 + Zn^{2+} \rightarrow H_3AsO_4 + Zn$$
 (acidic)

e.
$$CN^{1-} + MnO_4^{1-} \rightarrow CNO^{1-} + MnO_2$$
 (basic)

g.
$$ClO^{1-} + CrO_2^{1-} \rightarrow CrO_4^{2-} + Cl_2$$
 (basic)

Redox Reaction Exercise And Answers

Michal Rosen-Zvi

Redox Reaction Exercise And Answers:

Adopting the Beat of Term: An Psychological Symphony within Redox Reaction Exercise And Answers

In a global taken by screens and the ceaseless chatter of quick conversation, the melodic splendor and mental symphony developed by the written word usually disappear into the background, eclipsed by the constant sound and disturbances that permeate our lives. But, located within the pages of **Redox Reaction Exercise And Answers** a wonderful fictional prize brimming with fresh emotions, lies an immersive symphony waiting to be embraced. Constructed by an elegant composer of language, this charming masterpiece conducts readers on a psychological trip, well unraveling the hidden tunes and profound affect resonating within each cautiously crafted phrase. Within the depths of the moving evaluation, we shall examine the book is main harmonies, analyze its enthralling writing fashion, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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