Name	e:	Peri	od:			
Prope	erties of Water W	orksheet				
Word						
Covaler tension	and the second second second second second	ion dissolve condensation	cohesion negatively	surface adhesion	cooler positively	universal sublimation
1.	The hydrogen and oxyg	en atoms are hel	d together by	covalent		bends.
2.	The electrons are not s	hared equally cr	eating a <u>pol</u>	ar	mo	slecule.
3.	The polarity of water a	llows it todis	ssolve	most sul	ostances. E	Because of this
	it is referred to as the <u>universal</u>			solvent		
4.	Water molecules stick t	to other water m	olecules. This	property is	called	cohesion
5.	Hydrogen bonds form between adjacent water molecules because thepositively					
	charged hydrogen end of one water molecule attracts the negatively charged					
	oxygen end of another					
6.	oxygen end or anomer	vater malecule.				
6.	Water molecules stick t		ls due to its po	lar nature.	This prope	rty is called
6.	나는 보통 이번 사람들이 되는 것이 없었다. 그렇게 되었다.	to other materia	ls due to its po	dar nature.	This prope	rty is called
	Water molecules stick to adhesion	to other materia				
	Water molecules stick	to other materia				
7.	Water molecules stick to adhesion surface formed due to the polar	tension	۴.	_ creates ti	ne skin-liki	e surface
7.	Water molecules stick to adhesion surface formed due to the polar viscosity	tension tension nature of water	r. the tendency	_ creates the	ne skin-liki o resist flo	e surface
7.	Water molecules stick to adhesion surface. formed due to the polar viscosity cooler	tension renature of water is	r. the tendency more viscous	_ creates the	ne skin-liki o resist flo	e surface
7. 8.	Water molecules stick to adhesion surface formed due to the polar viscosity cooler organisms to use less en	tension nature of water is water is	r. the tendency more viscous tom sinking.	_ creates the control of the control	ne skin-like o resist flo water, all	e surface nw. owing floating
7. B.	Water molecules stick to adhesion surface formed due to the polar viscosity cooler organisms to use less er condensation	tension nature of water is water is	r. the tendency more viscous tom sinking. water changes	_ creates the control of the control	ne skin-like resist flo water, all to a liquid.	e surface nw. owing floating

12. Why does ice float?

water is less dense in solid form because hydrogen bonding between water molecules pushed molecules further apart into a set crystalline pattern

13. List the properties that ocean water has due to the fact that it contains salt.

higher density conductivity decreased surface tension

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