

Atomic Structure Review Worksheet

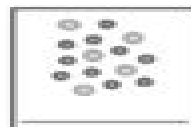
- ① Democritus of Abdera, who lived in Greece in the 4th century B.C., suggested that matter is made up tiny particles that cannot be divided. He called these particles _____.

- ② List two reasons why Democritus' ideas were not useful in a scientific sense.

① _____

② _____

- ③ Atoms of different element can physically mix together or chemically combine in simple whole-number ratios to form compounds. In the diagram below, use the labels mixture and compound to identify the mixture of elements A and B and the compound that forms when the atoms of element A and B combine chemically.



- ④ Complete the following sentences by underlining the correct words in parenthesis.
- a An element is a matter that is composed of one type of (atom/quark)
 - b The unit of measurement used for atomic particles is the (atom size/atomic mass unit)
 - c Atoms of the same element that have different number of neutrons are called (isotopes/electron cloud)
 - d In the periodic table, elements are arranged by increasing atomic (powder/number)
 - e An electron dot diagram uses the symbol of an element and dots to represents the (quarks/electrons) in the outer energy shell.
- ⑤ In your own words, state the main ideas of Dalton's atomic theory.

Regents Review Atomic Structure Answers

Albert S. Tarendash



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