#### Regents Chemistry Unit 8 Thermochemistry Practice Test

- At which temperature do both the solid and the liquid phase of water exist?
  - (1) 273 °C (2) 100 °C.

- (3) 0°C (4) 373 °C
- 2. What is the freezing point of sodium?
  - (I) 371°C (2) 644 °C

- (3) -98 °C (4) 98 °C
- 3. A 15.75-g piece of iron absorbs 1086.75 joules of heat energy, and its temperature changes from 25°C to 175°C. What is the specific heat capacity of iron?

  - (1) 1.0 J/g\*\*C (3) 0.46 J/g\*\*C
  - (2) 4.18 1/g+0C (4) 2.76 1/g+0C
- 4. Which statement concerning propanone and water at 50°C is true?
  - (1) Propanone has a lower vapor pressure and weaker intermolecular forces than water
  - (2) Propanine has a lower vapor pressure and stronger intermolecular forces than water.
  - (3) Propanone has a higher vapor pressure and weaker intermolecular forces than water.
  - (4) Propanone has a higher vapor pressure and stronger intermolecular forces than water.
- Which phase change is an exothermic process?
  - (I) CO<sub>2</sub> (s) → CO<sub>2</sub> (g)
- (2) Cia (s)--- Cu(A
- (3) NH<sub>3</sub>(I) → NH<sub>3</sub> (g)
- (4) Hg (A→ Hg (s)
- 6. What is the vapor pressure of water at 105°C?
  - (1) 100 kPa.

(3) 101.3 kPa

(2) 110 kPa

- (4) 120 kPa
- 7. As a substance changes from a liquid to a gas, the average distance between molecules
  - (II) decreases.
  - (2) increases
  - (3) remains the same

8. The graph below represents the uniform heating of a substance, starting with the substance as a solid below its melting point.



Which line segment represents an increase in average kinetic energy and so change in potential energy?

(I) BC

(3) ED

(2) AB

- (4) IDE:
- How many joules are required to melt 100, grams of water?
  - (1) 226,000 kJ
- (2)  $22,600 \pm$
- (3) 33,400 kJ
- (4) 33,400 [
- 10. What occurs when a 35-gram aluminum cube at 100,°C is place in 90, grams of water at 25°C in an insulated cup?
  - (1) Heat is transferred from the aluminum to the water, and the temperature of the water decreases.
  - (2) Heat is transferred from the aluminum to the water, and the temperare of the water increases.
  - (3) Heat is transferred from the water to the aluminum, and the temperature of the water decreases.
  - (4) Heat is transferred from the water to the aluminum, and the temperature of the water increases.
- 11. What is the boiling point of propanone if the pressure on its surface is 48 kilopascals?
  - (II) 25°C

(3) 35°C

(2) 30.°C

(4) 40.°C

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**Léo Pomar** 

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