

Electrochemistry Class 12 Important Questions Very Short Answer Type

Question 1.

What is meant by 'limiting molar conductivity'? (All India 2010)

Answer:

The molar conductivity of a solution at infinite dilution is called limiting molar conductivity and is represented by the symbol Λ_{∞} .

Question 2.

Express the relation between conductivity and molar conductivity of a solution held in a cell. (Delhi 2011)

Answer:

$\Lambda_{\infty} = \kappa C$ = Conductivity Concentration

Question 3.

What is the effect of catalyst on:

- (i) Gibbs energy (ΔG) and
- (ii) activation energy of a reaction? (Delhi 2017)

Answer:

- (i) There will be no effect of catalyst on Gibbs energy.
- (ii) The catalyst provides an alternative pathway by decreasing the activation energy of a reaction.

Question 4.

What is the effect of adding a catalyst on

- (a) Activation energy (E_a), and
- (b) Gibbs energy (ΔG) of a reaction? (All India 2017)

Answer:

- (a) On adding catalyst in a reaction, the activation energy reduces and rate of reaction is fastened.
- (b) A catalyst does not alter Gibbs energy (ΔG) of a reaction.

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Question 5.

Two half cell reactions of an electrochemical cell are given below :



Construct the redox equation from the two half cell reactions and predict if this reaction favours formation of reactants or product shown in the equation. (All India 2009)

Answer:

The reactions can be represented at anode and at cathode in the following ways :

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RC Schank



Redox And Electrochemistry Regents Short Answer Questions:

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In a digital era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Redox And Electrochemistry Regents Short Answer Questions**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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