## THE "OFFICIAL" CHEMISTRY 12 REDOX &ELECTROCHEMISTRY STUDY GUIDE

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#### INTRODUCTION

- 1. set Which one of the following is the strongest reducing agent?
- 2. Set In the reaction below:
  - $6H^+ + 6I^- + C1O_3^- \rightarrow 3I_2 + 3H_2O + CI^-$
- A. 1 2 B. F C. H (5) CIO 3 ..
- 3. 501 Which of the following is the strongest reducing agent?
  - A. cobult
  - B. copper calcium
- 4. 501 An oxidizing agent is

cheomium

- A. reduced as it loses electrons.
  reduced as it gains electrons.
  oxidized as it loses electrons.
  D. oxidized as it gains electrons.
- 5. 501 Which of the following is the strongest exidizing agent?

B. Mn<sup>2+</sup>
C acidified MnO<sub>2</sub>
D acidified MnO<sub>3</sub>

# **Redox Reaction Chem Study Guide Answers**

Marcel A. Müller

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