

Chem 12 Electrochemistry : Quiz 2 - 120

Part 1

1) A particular reaction has a predicted E° of -2.25 V. This indicates that the reaction would :

- a) occur rapidly b) produce energy c) go to completion
d) be non-spontaneous

2) Corrosion of a metal means that the metal :

- a) acts as a cathode b) undergoes oxidation
c) undergoes reduction d) acts as an oxidizing agent

3) Zinc metal is prepared by electrolysis as follows:



What is the minimum voltage necessary for the electrolysis reaction to proceed ?

- a) 2.12 b) 1.58 c) 1.96 d) 3.48

4) Which metal would provide the best cathodic protection for iron pipes ?

- a) Ag b) Sn c) Mg d) Cu

5) Which half-reaction best describes what is taking place at the anode during the electrolysis of a 1.0 M NaCl solution ?

- a) $\text{Na}^{+} + \text{e}^{-} \rightarrow \text{Na}$ b) $2\text{Cl}^{-} \rightarrow \text{Cl}_2(\text{g}) + 2\text{e}^{-}$
c) $2\text{H}^{+}(10^{-7}\text{M}) + 2\text{e}^{-} \rightarrow \text{H}_2(\text{g})$ d) $\text{H}_2\text{O} \rightarrow 1/2 \text{O}_2(\text{g}) + 2\text{H}^{+}(10^{-7}\text{M}) + 2\text{e}^{-}$

6) Which one of the following converts chemical energy to electrical energy ?

- a) Industrial production of aluminum b) Nickel plating
c) Discharging a "dry" cell d) Production of Na(s)

7) Which is the unit for the energy per coulomb of electrons ?

Redox And Electrochemistry Quiz Answers

NARAYAN CHANGDER



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In an electronic era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Redox And Electrochemistry Quiz Answers**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve to the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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