

Name: _____
Date: _____ Period: _____

Chemistry: *The Periodic Table*

Directions: Answer each of the following questions. You need not use complete sentences.

What is a period? How many are there in the periodic table?

A period is a row of elements on the periodic table, in which the elements are ordered one proton more each place from left to right. A period also often helps to represent an electron energy level and the order in which electrons are filled into their sublevels. There are 7 periods.

What is a group (also called a family)? How many are there in the periodic table?

Groups are the columns that are found top to bottom in the table. They show elements that have the same sublevel and number of electrons in their outer shell, valence electrons. Often elements in a group share any similar chemical properties because of this. There are 18 groups in the periodic table.

State the number of valence electrons in an atom of:

a. sulfur
6

b. calcium
2

c. chlorine
7

d. arsenic
5

Give the names and chemical symbols for the elements that correspond to these atomic numbers:

a. 10

Neon, Ne

b. 18

Argon, Ar

c. 36

Krypton, Kr

d. 90

Thorium, Th

List, by number, both the period and group of each of these elements.

	Symbol	Period	Group
a. beryllium	Be	2	2
b. iron	Fe	4	8
c. lead	Pb	6	14

Which of the following pairs of elements belong to the same period?

a. Na and Cl

b. Na and Li

c. Na and Cu

d. Na and Ne

Which of the following pairs of elements belong to the same group?

a. H and He

b. Li and Be

c. C and Pb

d. Ga and Ge

Would you expect strontium to be, chemically, more similar to calcium or rubidium and WHY?

Calcium, because they share a common amount of valence electrons. Valence electrons are the main contributor to chemical properties and how elements interact with other matter. Both strontium and calcium have two electrons on the the "s" sublevel, $5s^2$ and $4s^2$

What are the Group 1 elements called?

Periodic Table Assessment Key Answers

IM Harris



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Table of Contents Periodic Table Assessment Key Answers

1. Understanding the eBook Periodic Table Assessment Key Answers
 - The Rise of Digital Reading Periodic Table Assessment Key Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Assessment Key Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Assessment Key Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Assessment Key Answers

- Personalized Recommendations
- Periodic Table Assessment Key Answers User Reviews and Ratings
- Periodic Table Assessment Key Answers and Bestseller Lists
- 5. Accessing Periodic Table Assessment Key Answers Free and Paid eBooks
 - Periodic Table Assessment Key Answers Public Domain eBooks
 - Periodic Table Assessment Key Answers eBook Subscription Services
 - Periodic Table Assessment Key Answers Budget-Friendly Options
- 6. Navigating Periodic Table Assessment Key Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Table Assessment Key Answers Compatibility with Devices
 - Periodic Table Assessment Key Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table Assessment Key Answers
 - Highlighting and Note-Taking Periodic Table Assessment Key Answers
 - Interactive Elements Periodic Table Assessment Key Answers
- 8. Staying Engaged with Periodic Table Assessment Key Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Assessment Key Answers
- 9. Balancing eBooks and Physical Books Periodic Table Assessment Key Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Table Assessment Key Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Table Assessment Key Answers
 - Setting Reading Goals Periodic Table Assessment Key Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Table Assessment Key Answers

- Fact-Checking eBook Content of Periodic Table Assessment Key Answers
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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