

Equilibrium and Reaction Rates

(answers at the end)

Q: What is Le Chatelier's law (Le Chatelier's principle)?

A: Study Le Chatelier's principle's meaning.

A: Understanding that when the system is subjected to a perturbation, it reacts.

A: Study Le Chatelier's law, which predicts the direction of the reaction when the system is subjected to a perturbation (change in concentration, pressure, temperature).

A: Le Chatelier's law predicts the direction of the reaction when the system is subjected to a perturbation.

A: Le Chatelier's law predicts the direction of the reaction when the system is subjected to a perturbation. The direction of the reaction is determined by the sign of the reaction quotient, which is compared to the equilibrium constant.

Q: What is the reaction quotient (Q) and how is it calculated?

A: The reaction quotient (Q) is calculated as follows:

$$Q = \frac{[C]^c [D]^d}{[A]^a [B]^b} \quad \text{for } aA + bB \rightleftharpoons cC + dD$$

$$Q = \frac{[C]^c [D]^d}{[A]^a [B]^b} \quad \text{for } aA + bB \rightleftharpoons cC + dD$$

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Compare these equations to those of the equilibrium constant (K) to determine the direction of the reaction.

A: What is the reaction quotient?

Q: What is the reaction quotient (Q) and how is it calculated?

- (A) $Q = \frac{[C]^c [D]^d}{[A]^a [B]^b}$
- (B) $Q = \frac{[A]^a [B]^b}{[C]^c [D]^d}$
- (C) $Q = \frac{[C]^c [D]^d}{[A]^a [B]^b}$
- (D) It depends on the initial concentrations.
- (E) It is calculated at equilibrium.

A: What is the reaction quotient (Q) and how is it calculated?

A: The reaction quotient (Q) is calculated as follows:

Reactions Rates And Equilibrium Section Review Answers

Ralph Thomas Myers



Reactions Rates And Equilibrium Section Review Answers:

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Systems and Materials Nicolas Giuseppone, Andreas Walther, 2021-07-19 Out of Equilibrium Supra molecular Systems and Materials A must have resource that covers everything from out of equilibrium chemical systems to active materials Out of

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