

- Q6 When water was added to the anhydrate of $\text{CuSO}_4(x)$, did the reaction proceed in the direction of the reactants or products?

B.2 Reaction of Copper(II) Ion and Hydroxide

	Test tube 1	Test tube 2
Initial color of CuCl_2 solution (1)	light blue	light blue
Initial appearance $\text{Cu(OH)}_2(x)$ (2)	clear	clear
Change in appearance (3)	cloudy	clear
Change in appearance (4)	cloudy	clear

- Q7 In test tube 1, did the addition of OH^- cause the reaction to go in the direction of the reactants or products? Explain.
- Q8 In test tube 2, did the addition of H_3O^+ cause the reaction to go in the direction of the reactants or products? Explain.

C. Changing Equilibrium Conditions: Le Châtelier's Principle

C.1 Effect of Changing Reactant Concentration

Test tube	Color	Is color deeper or lighter red compared to the reference?	$[\text{FeSCN}^{2+}]$ increases or decreases?
1. Reference	red-orange	deeper/lighter	increases/decreases
2. Fe^{3+} added	dark red	deeper	increased
3. SCN^- added	blood red	deeper	increased
4. Fe^{3+} removed	orange	lighter	decreased

Reaction Rates And Equilibrium Lab Answers

Peipei Pang



Reaction Rates And Equilibrium Lab Answers:

Reaction Rates for High-temperature Air with Carbon and Sodium Impurities Mina L. Carnicom, 1968 The values used by a number of investigators for the rate constants of high temperature greater than or equal to 1000 K homogeneous gaseous reactions involving species of the elements nitrogen oxygen carbon and sodium have been compiled and are presented in tabular form Included are reactions involving neutral species charged species free electrons some species in excited electronic or vibrational states and radiative processes Safety-Scale Lab Exp Biochem 2e Spencer L.

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emphasis on handling data is supported by the Data Analysis section Geochemical Modeling for Mine Site Characterization and Remediation D. Kirk Nordstrom, Andrew Nicholson, 2017-10-01 The single most important factor for the successful application of a geochemical model is the knowledge and experience of the individual s conducting the modeling Geochemical Modeling for Mine Site Characterization and Remediation is the fourth of six volumes in the Management Technologies for Metal Mining Influenced Water series about technologies for management of metal mine and metallurgical process drainage This handbook describes the important components of hydrogeochemical modeling for mine environments primarily those mines where sulfide minerals are present metal mines and coal mines It provides general guidelines on the strengths and limitations of geochemical modeling and an overview of its application to the hydrogeochemistry of both unmined mineralized sites and those contaminated from mineral extraction and mineral processing The handbook includes an overview of the models behind the codes explains vital geochemical computations describes several modeling processes provides a compilation of codes and gives examples of their application including both successes and failures Hydrologic modeling is also included because mining contaminants most often migrate by surface water and groundwater transport and contaminant concentrations are a function of water residence time as well as pathways This is an indispensable resource for mine planners and engineers environmental managers land managers consultants researchers government regulators nongovernmental organizations students stakeholders and anyone with an interest in mining influenced water The other handbooks in the series are Basics of Metal Mining Influenced Water Mitigation of Metal Mining Influenced Water Mine Pit Lakes Characteristics Predictive Modeling and Sustainability Techniques for Predicting Metal Mining Influenced Water and Sampling and Monitoring for the Mine Life Cycle Chemistry McGraw-Hill Staff, 2001-07 *ERDA Energy Research Abstracts* ,1983 **Subject Index to Unclassified ASTIA Documents** Defense Documentation Center (U.S.), 1960 Solar Energy Update ,1982 **The Software Encyclopedia 2001** ,2001 **Government Reports Announcements & Index** ,1983 Aero/space Engineering ,1958

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