

EXPERIMENT 9 **CHEMICAL REACTIONS: STOICHIOMETRY** **LABORATORY REPORT**

Your Name _____ TA's Name _____ Lab Section 1411. _____ Date _____

You must read pages 69 and 70 to complete the following calculations.

	Trial #1	Trial #2	Trial #3
Vol. 0.100 M K_2CrO_4 solution	<u>12.00 mL</u>	<u>12.00 mL</u>	<u>12.00 mL</u>
mmole K_2CrO_4	<u>1.20</u>		
Vol. 0.100 M $Pb(NO_3)_2$ solution	<u>10.00 mL</u>	<u>12.00 mL</u>	<u>14.00 mL</u>
mmole $Pb(NO_3)_2$	<u>1.0</u>		
mass product + paper	<u>0.579 g</u>	<u>0.654 g</u>	<u>0.639 g</u>
mass dry filter paper	<u>0.256 g</u>	<u>0.266 g</u>	<u>0.251 g</u>
mass $PbCrO_4$ collected precip.	<u>.323</u>		
mmole $PbCrO_4$	<u>1.0 mmole</u>	<u>1.2</u>	

CALCULATIONS AND QUESTIONS

(USE THE SPACE ON THE BACK OF THIS SHEET IF NECESSARY)

1. In each trial determine which reagent was the limiting reagent? Give a reason, determine the results of your experiment, which proves your conclusion.



2. Calculate the number of millimoles of K_2CrO_4 in 10.55 mL of a 0.120 M solution. Report answer with 3 significant figures (SF).

Reaction Stoichiometry Lab Answers

Lei Shi

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name 'Lei Shi'.

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Table of Contents Reaction Stoichiometry Lab Answers

1. Understanding the eBook Reaction Stoichiometry Lab Answers
 - The Rise of Digital Reading Reaction Stoichiometry Lab Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Reaction Stoichiometry Lab Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Reaction Stoichiometry Lab Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Reaction Stoichiometry Lab Answers

- Personalized Recommendations
- Reaction Stoichiometry Lab Answers User Reviews and Ratings
- Reaction Stoichiometry Lab Answers and Bestseller Lists
- 5. Accessing Reaction Stoichiometry Lab Answers Free and Paid eBooks
 - Reaction Stoichiometry Lab Answers Public Domain eBooks
 - Reaction Stoichiometry Lab Answers eBook Subscription Services
 - Reaction Stoichiometry Lab Answers Budget-Friendly Options
- 6. Navigating Reaction Stoichiometry Lab Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Reaction Stoichiometry Lab Answers Compatibility with Devices
 - Reaction Stoichiometry Lab Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Reaction Stoichiometry Lab Answers
 - Highlighting and Note-Taking Reaction Stoichiometry Lab Answers
 - Interactive Elements Reaction Stoichiometry Lab Answers
- 8. Staying Engaged with Reaction Stoichiometry Lab Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Reaction Stoichiometry Lab Answers
- 9. Balancing eBooks and Physical Books Reaction Stoichiometry Lab Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Reaction Stoichiometry Lab Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Reaction Stoichiometry Lab Answers
 - Setting Reading Goals Reaction Stoichiometry Lab Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Reaction Stoichiometry Lab Answers

- Fact-Checking eBook Content of Reaction Stoichiometry Lab Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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