

TUTORIAL 3

ALKENES & CYCLOALKENES

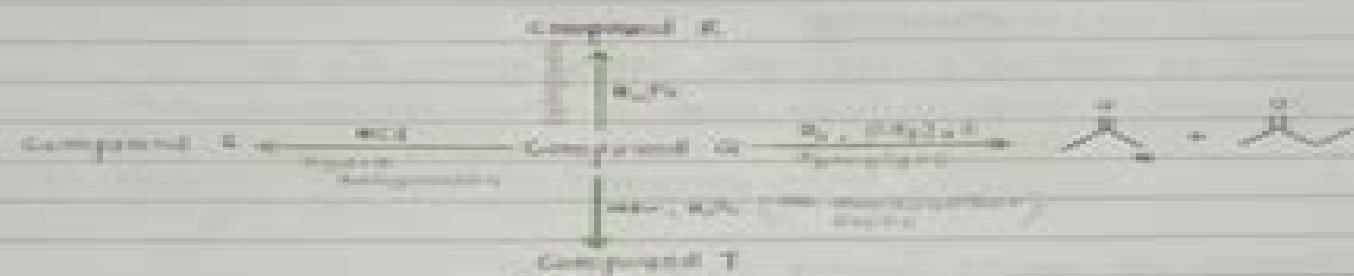
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Q1

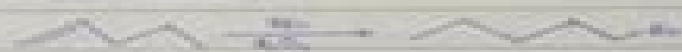
Alkene can undergo various reactions. Identify the structure of R, S, T and U.



Compound R (alkene)



Compound T (anti-oxidation reagent)



Compound S (hydrogenation)



Compound U (hydrohalogenation)



Q2

a) Define hydrocarbon compounds

A class of organic compounds made up of only the elements carbon (C) and hydrogen (H)

b) Indicate the difference between propene and propane

Propene	Propane
- Propene is a saturated hydrocarbon compound	- Propane is a unsaturated hydrocarbon compound
- carbon-carbon double bond	- carbon-carbon single bond in the structure
- fully saturated to bond	- double bond and single bond

Reaction Lab Answer Key Hydrocarbon

M Walker

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the 'M Walker' text.

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Table of Contents Reaction Lab Answer Key Hydrocarbon

1. Understanding the eBook Reaction Lab Answer Key Hydrocarbon
 - The Rise of Digital Reading Reaction Lab Answer Key Hydrocarbon
 - Advantages of eBooks Over Traditional Books
2. Identifying Reaction Lab Answer Key Hydrocarbon
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Reaction Lab Answer Key Hydrocarbon
 - User-Friendly Interface
4. Exploring eBook Recommendations from Reaction Lab Answer Key Hydrocarbon
 - Personalized Recommendations
 - Reaction Lab Answer Key Hydrocarbon User Reviews and Ratings
 - Reaction Lab Answer Key Hydrocarbon and Bestseller Lists

5. Accessing Reaction Lab Answer Key Hydrocarbon Free and Paid eBooks
 - Reaction Lab Answer Key Hydrocarbon Public Domain eBooks
 - Reaction Lab Answer Key Hydrocarbon eBook Subscription Services
 - Reaction Lab Answer Key Hydrocarbon Budget-Friendly Options
6. Navigating Reaction Lab Answer Key Hydrocarbon eBook Formats
 - ePub, PDF, MOBI, and More
 - Reaction Lab Answer Key Hydrocarbon Compatibility with Devices
 - Reaction Lab Answer Key Hydrocarbon Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Reaction Lab Answer Key Hydrocarbon
 - Highlighting and Note-Taking Reaction Lab Answer Key Hydrocarbon
 - Interactive Elements Reaction Lab Answer Key Hydrocarbon
8. Staying Engaged with Reaction Lab Answer Key Hydrocarbon
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Reaction Lab Answer Key Hydrocarbon
9. Balancing eBooks and Physical Books Reaction Lab Answer Key Hydrocarbon
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Reaction Lab Answer Key Hydrocarbon
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Reaction Lab Answer Key Hydrocarbon
 - Setting Reading Goals Reaction Lab Answer Key Hydrocarbon
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Reaction Lab Answer Key Hydrocarbon
 - Fact-Checking eBook Content of Reaction Lab Answer Key Hydrocarbon
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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