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b) propyne c) ethyne a) Butyne d) pentyne

2) The loss of _____ and ____ atoms from the double-bonded carbon atoms leads to alkyne formation.

 a) Oxygen, halogen
b) oxygen, hydrogen
c) hydrogen, d) carbon, halogen halogen

What is the reagent used and appropriate temperature required to prepare ethene from ethyl alcohol?

a) Diluted H₂SO₄, 160°C

b) Concentrated H₂SO₄, 170°C

c) Diluted HCl, 170°C

d) Concentrated HCl, 160°C

4) _____ on heating with alcoholic potash loses one molecule of hydrogen halide to give alkene.

a) Halogen

b) Haloalkanes

c) Haloalkynes

d) None of the above

5) The IUPAC name of the compound H2C=CH-CH(CH3)2:

a) 2-Vinylpropene

b) 1,1-dimethyl-2-peopene

c) 3-Methyl-1-butene

d) 1-Isopropyl ethylene

6) Ethylbenzene reacts with Cl₂ in sunlight to give:

a) I-Chloro-1-phenylethane

b) 1-Chloro-2-phenylethane

c) 4-Chloroethylbenzene

d) 3-Chloroethylbenzene

7) Benzene Can be :

a) Discharge the color of KMNO₄

b) Discharge the color of bromine water

c) Polymerize

d) Undergo substitution reactions

8) Which of the following reactions will yield 2, 2 -dibromopropane?

a) HC == CH + 2HBr -->

b) H₃C — ⊆ = CH₂ + HB_F — ►

c) H₃C — C == CHBr + 2HBr — ► d) H₃C — C == CHBr + HBr — ►

Give the structure of reactants :

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- 1. Understanding the eBook Quiz And Answers About Organic Chemistry
 - The Rise of Digital Reading Quiz And Answers About Organic Chemistry
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Quiz And Answers About Organic Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Quiz And Answers About Organic Chemistry
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Quiz And Answers About Organic Chemistry
 - Personalized Recommendations
 - Quiz And Answers About Organic Chemistry User Reviews and Ratings
 - Quiz And Answers About Organic Chemistry and Bestseller Lists
- 5. Accessing Quiz And Answers About Organic Chemistry Free and Paid eBooks
 - Quiz And Answers About Organic Chemistry Public Domain eBooks
 - Quiz And Answers About Organic Chemistry eBook Subscription Services
 - Quiz And Answers About Organic Chemistry Budget-Friendly Options
- 6. Navigating Quiz And Answers About Organic Chemistry eBook Formats

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- Quiz And Answers About Organic Chemistry Compatibility with Devices
- Quiz And Answers About Organic Chemistry Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Quiz And Answers About Organic Chemistry
 - Highlighting and Note-Taking Quiz And Answers About Organic Chemistry
 - Interactive Elements Quiz And Answers About Organic Chemistry
- 8. Staying Engaged with Quiz And Answers About Organic Chemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Quiz And Answers About Organic Chemistry
- 9. Balancing eBooks and Physical Books Quiz And Answers About Organic Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Quiz And Answers About Organic Chemistry
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Quiz And Answers About Organic Chemistry
 - Setting Reading Goals Quiz And Answers About Organic Chemistry
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Quiz And Answers About Organic Chemistry
 - Fact-Checking eBook Content of Quiz And Answers About Organic Chemistry
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
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