

## Physical Science 4 Study Guide

- Physical and Chemical Properties

1. What is the difference between a physical change and a chemical change?

*For a physical change, there is no change in chemical composition. What you start out with is the same as what you end up with.*

*For a chemical change there is a change in chemical composition. What you start out with is different than what you end up with.*

2. Which type of property involves a change in the chemical structure?

*Chemical property*

3. Which type of property is there no change in the chemical structure?

*Physical property*

4. Which type of property do you end up with something different than what you started with?

*Chemical property*

5. Which type of property do you end up with the same thing as what you started with?

*Physical property*

6. What are some examples of physical properties?

*Solubility*

*Boiling Point*

*Freezing/Melting Point*

*Viscosity*

*Electrical Conductivity*

*Density*

7. What are some examples of chemical properties?

*Corrosion*

*Oxidation*

*Combustion*

8. Does the term boiling point refer to a temperature or a phase change?

*Temperature*

9. Does the composition of a substance change during a phase change?

*No.*

10. 5.0g of substance X has a volume of 10.0 mL. What is the density of substance X?

*$5.0\text{ g} / 10.0\text{ mL} = .5\text{ g/mL}$*

11. Is the density of solid substance X the same as liquid substance X?

*No.*

12. While making Koolaid you add some sugar to some water. In this solution, sugar is the (solute, solvent) and water is the (solute, solvent).

# Physical Science 4 Study Guide Answers

**Smithsonian Institution, National  
Academy of Engineering, National  
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