

TYBSC SEM VI PHYSICAL CHEMISTRY (MODEL MCQ's)

|   |                                                                                                                                                                                                                                                                                                                                                                                                                                                |
|---|------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|
| 1 | Which of the following is a uni-univalent electrolyte?<br>a) KCl<br>b) $\text{CuCl}_2$<br>c) $\text{AlCl}_3$<br>d) $\text{MgCl}_2$                                                                                                                                                                                                                                                                                                             |
| 2 | The ionic strength of a solution of molality $m_i$ and valency of the ion $Z_i$ is ____<br>a) $\mu = \frac{1}{2} m_i Z_i^2$<br>b) $\mu = \sum m_i Z_i^2$<br>c) $\mu = \frac{1}{2} \sum Z_i^2$<br>d) $\mu = \frac{1}{2} \sum m_i Z_i^2$                                                                                                                                                                                                         |
| 3 | Which of the following is an example of electrolyte concentration cell with transference?<br><br>a) $\text{Cu}   \text{CuSO}_4    \text{AgNO}_3   \text{Ag}$<br>b) $\text{Pt}, \text{H}_{2(g)}   \text{HCl}_{(aq)}   \text{AgCl}   \text{Ag}$<br>c) $\text{Pt}, \text{H}_{2(g)}   \text{HCl}_{(aq)}(a_1)   \text{AgNO}_3(a_2)   \text{Ag}$<br>d) $\text{Ag} - \text{AgCl}   \text{NaCl}(a_1)   \text{NaCl}(a_2)   \text{AgCl} \cdot \text{Ag}$ |
| 4 | A concentration cell which has salt bridge in its representation is ____<br>a) electrolyte concentration cell with transference<br>b) Electrode concentration cell<br>c) Electrolyte concentration cell without transference<br>d) Gas concentration cell                                                                                                                                                                                      |
| 5 | The total emf of the electrolyte concentration cell with transference containing the liquid junction potential is given by.....<br>a) $E_j = E_{\text{oxidation}} + E_{\text{reduction}}$<br>b) $E_{\text{cell}} = E_{\text{oxidation}} + E_{\text{reduction}} + E_j$<br>c) $E_{\text{cell}} = E_{\text{oxidation}} - E_{\text{reduction}} - E_j$<br>d) $E_j = E_{\text{oxidation}} - E_{\text{reduction}}$                                    |
| 6 | The activity 'a' is related to the activity coefficient ' $\gamma$ ' by the equation____<br>a) $a = \frac{m}{\gamma}$<br>b) $\gamma = a - m$<br>c) $a = m + \gamma$<br>d) $a = m\gamma$                                                                                                                                                                                                                                                        |

# Physical Chemistry Multiple Choice Questions

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**Physical Chemistry Multiple Choice Questions:**

## **Physical Chemistry Multiple Choice Questions** Book Review: Unveiling the Power of Words

In a global driven by information and connectivity, the power of words has be much more evident than ever. They have the ability to inspire, provoke, and ignite change. Such is the essence of the book **Physical Chemistry Multiple Choice Questions**, a literary masterpiece that delves deep to the significance of words and their impact on our lives. Compiled by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall affect readers.

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