

Q9. Draw the bond line structure of heptan-4-one.

Q10. Draw the structure and IUPAC name of an alkane having the lowest molecular mass containing a quaternary carbon.

Q11. Answer the following questions.

- How many sigma and pi bonds are present in ethane?
- What is the hybridisation of all the carbons in propene?

Q12. What type of structural isomerism is shown by



Q13. Answer the following questions.

- Name a compound that does not contain halogen but gives a positive Beilstein's test.
- What will happen during Lassaignes's test for nitrogen if sulphur impurities are present in the compound?

Q14. How will you separate a mixture of ortho nitrophenol and para nitrophenol?

Q15. How will you separate benzoic acid from a mixture of benzoic acid and naphthalene?

Q16. What are electrophiles and nucleophiles? Give some examples.

Q17. Draw the condensed and bond line formula of the following compounds.



Q18. Suggest methods for the separation of the following mixtures.

- A mixture of liquid A (Boiling point = 365 K) and liquid B (Boiling point = 355 K).
- A mixture of liquid A (Boiling point = 348 K) and liquid B (Boiling point = 578 K).

Q19. Answer the following questions.

- What is the cause of geometrical isomerism?
- Why do we fuse organic compounds with sodium while testing for halogen, nitrogen, sulphur and phosphorus?

Q20. Differentiate between inductive and resonance effects.

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