

Quiz 1: Equilibria, Solids & Gases

Multiple Choice

Answer the lower of the choices that best completes the statement or answers the question.

CONTENT REVIEW

- _____ 1. Chemical equilibrium exists when
- the processes cease.
 - the driving forward processes cease.
 - the opposing processes occur at different rates.
 - the opposing processes occur at the same rate.
- _____ 2. Any chemical reaction in which the products are represented by brackets is called a(n)
- double-replacement reaction.
 - reversible reaction.
 - single-replacement reaction.
 - precipitation reaction.
- _____ 3. What symbol is used to indicate a reversible reaction?
- $\rightleftharpoons$
 - $\rightleftarrows$
 - $\rightleftharpoons$
 - $\rightleftarrows$
- _____ 4. What is the general relationship between the concentrations and the rate of a forward reaction?
- Rate is greater when product concentration is greater.
 - Rate is greater when reactant concentration is greater.
 - Rate is greater when catalyst concentration is lower.
 - No relationship exists between concentrations and rate.
- _____ 5. What symbol is used to denote concentration?
- $[]$
 - $()$
 - $\{ \}$
 - $\langle \rangle$
- _____ 6. At chemical equilibrium, which of the following is always true of the concentrations of reactants and products?
- They are constant.
 - Reactant concentration is zero.
 - They are equal.
 - Product concentration is zero.
- _____ 7. What is the equilibrium constant for the general reaction $aA + bB \rightleftharpoons cC + dD$?
- $\frac{[A]^a[B]^b}{[C]^c[D]^d}$
 - $\frac{[C]^c[D]^d}{[A]^a[B]^b}$
 - $\frac{[C]^c[D]^d}{[A]^a[B]^b}$
 - $\frac{[C]^c[D]^d}{[A]^a[B]^b}$
- _____ 8. The symbol for the equilibrium constant is
- K'
 - K_{eq}
 - K_{eq}
 - K'
- _____ 9. The principle that relates changes in pressure on equilibrium systems to equilibrium position is
- Boyle's law.
 - Le Chatelier's principle.
 - the law of chemical equilibrium.
 - Avogadro's principle.
- _____ 10. If a change in conditions is imposed on a system at equilibrium, the equilibrium position will always

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