# **Balancing Chemical Equations Practice**

Balance the following chemical equations.

1. Fe + 
$$O_2 \rightarrow$$
 Fe<sub>3</sub>O<sub>4</sub>

3. 
$$Sn + NaOH \rightarrow Na_2SnO_2 + H_2$$

4. 
$$K + Br_2 \rightarrow KBr$$

5. 
$$C_8H_{18} + O_2 \rightarrow CO_2 + H_2O$$

7. 
$$COCl_2 + H_2O \rightarrow HCl + CO_2$$

8. 
$$CS_2 + O_2 \rightarrow CO_2 + SO_2$$

9. 
$$H_2SO_4 + NaCN \rightarrow HCN + Na_2SO_4$$

11. 
$$H_2 + F_2 \rightarrow HF$$

12. BaCl<sub>2</sub> + KIO<sub>3</sub> 
$$\rightarrow$$
 Ba(IO<sub>3</sub>)<sub>2</sub> + KCl

13. 
$$Mg + HCl \rightarrow MgCl_2 + H_2$$

# **Quiz Balancing Chemical Equations 10 Questions**

**L** Cohen

## **Quiz Balancing Chemical Equations 10 Questions:**

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web step 2 the single loop secures the beads in place now it is a dangle the next and final step add the finding to convert the dangle depending on the finding will determine what the item becomes since we are talking basic beaded earrings we add a fish hook style earring wire to the dangle to make the earring

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