

# LAB: CALCULATING A LIMITING REACTANT

Name \_\_\_\_\_

## Lab Determination of a Limiting Reactant

**Objective:** Observe chemical reaction between magnesium metal and hydrochloric acid. Calculate the limiting and excess reactant.

**Materials:** 2 M HCl test tube

### Procedure:

1. Record the mass of magnesium: \_\_\_\_\_ g
2. Place the test tube in a test tube rack. Add the magnesium strip.
3. Add 4 mL of 2 M HCl into the test tube.
4. Record your observations. How do you know a chemical reaction is taking place? Be sure to feel the side of the test tube.
5. Pour the contents into the waste beaker provided by your teacher. Rinse the rack upside down. You never want to pour unreacted solid.
6. Write the balanced equation for the reaction that you observed and balance the equation.



# Quick Lab Limiting Reagents

**Scott C. Dulebohn**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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neutrons to get an answer of 9 2 the lowest principal quantum number for an electron is 0 1 2 3

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