

(iii) Suggest why the mean bond enthalpy of a C=C bond is less than twice the mean bond enthalpy of a C—C bond.

(1)

(iv) Use the mean bond enthalpy data in the table, and the equation given below, to calculate a value for the standard enthalpy change of combustion of propene.

(3)

Bond	Mean bond enthalpy / kJ mol ⁻¹
C=C	612
C—C	347
C—H	413
O=O	498
C=O	805
O—H	464



Answer = _____ kJ mol⁻¹



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1. Understanding the eBook Question Paper Of Edexcel M1 Jan 2015
 - The Rise of Digital Reading Question Paper Of Edexcel M1 Jan 2015
 - Advantages of eBooks Over Traditional Books
2. Identifying Question Paper Of Edexcel M1 Jan 2015
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Question Paper Of Edexcel M1 Jan 2015
 - User-Friendly Interface
4. Exploring eBook Recommendations from Question Paper Of Edexcel M1 Jan 2015

- Personalized Recommendations
 - Question Paper Of Edexcel M1 Jan 2015 User Reviews and Ratings
 - Question Paper Of Edexcel M1 Jan 2015 and Bestseller Lists
5. Accessing Question Paper Of Edexcel M1 Jan 2015 Free and Paid eBooks
- Question Paper Of Edexcel M1 Jan 2015 Public Domain eBooks
 - Question Paper Of Edexcel M1 Jan 2015 eBook Subscription Services
 - Question Paper Of Edexcel M1 Jan 2015 Budget-Friendly Options
6. Navigating Question Paper Of Edexcel M1 Jan 2015 eBook Formats
- ePub, PDF, MOBI, and More
 - Question Paper Of Edexcel M1 Jan 2015 Compatibility with Devices
 - Question Paper Of Edexcel M1 Jan 2015 Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Question Paper Of Edexcel M1 Jan 2015
 - Highlighting and Note-Taking Question Paper Of Edexcel M1 Jan 2015
 - Interactive Elements Question Paper Of Edexcel M1 Jan 2015
8. Staying Engaged with Question Paper Of Edexcel M1 Jan 2015
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Question Paper Of Edexcel M1 Jan 2015
9. Balancing eBooks and Physical Books Question Paper Of Edexcel M1 Jan 2015
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Question Paper Of Edexcel M1 Jan 2015
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Question Paper Of Edexcel M1 Jan 2015
- Setting Reading Goals Question Paper Of Edexcel M1 Jan 2015
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Question Paper Of Edexcel M1 Jan 2015

- Fact-Checking eBook Content of Question Paper Of Edexcel M1 Jan 2015
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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