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Economics Grade 11 Term 3 Question Paper and Memo for 2025

QUESTION 1: PROBLEM-SOLVING AND CREATIVE THINKING

1.1

1.1.1 Define the terms **problem-solving** and **decision-making** in your own words.

(4)

1.1.2 Give **four differences** between problem-solving and decision-making in a business context.

(4)

1.2 Read the scenario below and answer the questions that follow.

LATHA CLOTHING MANUFACTURERS (LCM)

Latha Clothing Manufacturers is struggling to meet its production targets. The machines used for sewing are outdated and frequently break down. Some workers suggested using local manufacturing tech companies for parts and assistance. The owner is not sure how to manage this new idea, but also wants to avoid more delays.

1.2.1 Identify a **creative thinking technique** that LCM can use to solve its problems. Quote from the scenario to support your answer.

(2)

1.2.2 Describe **three steps** LCM should take when applying the technique mentioned in QUESTION 1.2.1.

(6)

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Xiang Xie



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web october 3 2017 abstract the techniques used in this lab are quantitative transfer and vacuum filtration with the reaction of 8 001 grams of copper ii sulfate CuSO_4 and 2 0153 grams of iron powder Fe the goal of this experiment was to determine the product of copper ii sulfate with iron

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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8 00 grams of copper ii sulfate onto a piece of filter paper taking the filter paper s mass into account add these to the beaker 3 measure 50 0 ml of water in the cylinder and add it to the beaker 4

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web description great lab for use during stoichiometry unit in chemistry reinforces ideas of mole ratios and stoichiometry calculations by measuring the mass of iron that reacts and the mass of copper metal produced students can calculate the ratio of moles of reactant to moles of product this experimental mole ratio from lab can be compared

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web in this experiment we will use stoichiometric principles to deduce the appropriate equation for the reaction between metallic iron and a solution of copper ii sulfate this reaction produces metallic copper which is seen precipitating as a finely divided red powder

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web this experiment is focused on a reaction between iron metal and a solution of copper ii sulfate we are able to figure out the chemical equation between the two by using stoichiometric principles the iron displaces the copper ii sulfate ions the copper ii is then reduced and gains electrons

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web stoichiometry can be defined as the relationship between the relative quantities of substances taking part in a reaction or forming a compound stoichiometric principles can be used to deduce the appropriate chemical equation for a reaction in our experiment we used solid iron in the form of powder was added to a solution of copper ii sulfate

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web under the finalize of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7 0002g away cooper ii sulfate and 2 0101g iron metal which produced 2 3037g of copper product with a percent yield of 100 72 introduction

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