

ICSE Board
Class X Chemistry
Board Paper – 2012

Time: 1½ hrs

Total Marks: 80

General Instructions:

1. Answers to this paper must be written on the paper provided separately.
2. You will NOT be allowed to write during the first 15 minutes. This time is to be spent in reading the question paper.
The time given at the head of this paper is the time allowed for writing the answers.
This question paper is divided into two sections.
3. **Section I** contains one question with parts (a) to (h); all the eight parts are to be answered.
4. **Section II** contains six questions numbered 2 to 7. You are to answer any four of these questions.
The intended marks of questions or for parts of questions are given in brackets [].

SECTION I (40 Marks)

Attempt all questions from this section.

Question 1

(a) Name the gas in each of the following:

[5]

- i. The gas evolved on reaction of aluminium with boiling concentrated caustic alkali solution.
- ii. The gas produced when excess ammonia reacts with chlorine.
- iii. A gas which turns acidified potassium dichromate clear green.
- iv. The gas produced when copper reacts with concentrated nitric acid.
- v. The gas produced on reaction of dilute sulphuric acid with a metallic sulphide.

(b) State one observation for each of the following:

[5]

- i. Excess ammonium hydroxide solution is added to lead nitrate solution.
- ii. Bromine vapours are passed into a solution of ethyne in carbon tetrachloride.
- iii. A zinc granule is added to copper sulphate solution.
- iv. Zinc nitrate crystals are strongly heated.
- v. Sodium hydroxide solution is added to ferric chloride solution at first a little and then in excess.

Question Paper 2011 Icse Board Of Chemistry

Michael Seilmaier

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Table of Contents Question Paper 2011 Icse Board Of Chemistry

1. Understanding the eBook Question Paper 2011 Icse Board Of Chemistry
 - The Rise of Digital Reading Question Paper 2011 Icse Board Of Chemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying Question Paper 2011 Icse Board Of Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Question Paper 2011 Icse Board Of Chemistry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Question Paper 2011 Icse Board Of Chemistry
 - Personalized Recommendations
 - Question Paper 2011 Icse Board Of Chemistry User Reviews and Ratings

- Question Paper 2011 Icse Board Of Chemistry and Bestseller Lists
- 5. Accessing Question Paper 2011 Icse Board Of Chemistry Free and Paid eBooks
 - Question Paper 2011 Icse Board Of Chemistry Public Domain eBooks
 - Question Paper 2011 Icse Board Of Chemistry eBook Subscription Services
 - Question Paper 2011 Icse Board Of Chemistry Budget-Friendly Options
- 6. Navigating Question Paper 2011 Icse Board Of Chemistry eBook Formats
 - ePub, PDF, MOBI, and More
 - Question Paper 2011 Icse Board Of Chemistry Compatibility with Devices
 - Question Paper 2011 Icse Board Of Chemistry Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Question Paper 2011 Icse Board Of Chemistry
 - Highlighting and Note-Taking Question Paper 2011 Icse Board Of Chemistry
 - Interactive Elements Question Paper 2011 Icse Board Of Chemistry
- 8. Staying Engaged with Question Paper 2011 Icse Board Of Chemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Question Paper 2011 Icse Board Of Chemistry
- 9. Balancing eBooks and Physical Books Question Paper 2011 Icse Board Of Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Question Paper 2011 Icse Board Of Chemistry
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Question Paper 2011 Icse Board Of Chemistry
 - Setting Reading Goals Question Paper 2011 Icse Board Of Chemistry
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Question Paper 2011 Icse Board Of Chemistry
 - Fact-Checking eBook Content of Question Paper 2011 Icse Board Of Chemistry
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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