

# Chemistry Unit Test #5 SNC2D Name:

\_\_\_\_K \_\_\_\_T \_\_\_\_C \_\_\_\_A  
(Product)

Assessment of Learning

15 15 10 10

Multiple Choice: Circle the most correct answer. (15 P)

1. The symbol for the element potassium is \_\_\_\_\_.  
A) K B) P C) K D) S E) Ca

2. Which among the following is NOT both a molecule and a compound?  
A)  $\text{CaH}_2\text{O}_4$  B)  $\text{H}_2\text{O}$  C)  $\text{CO}_2$  D)  $\text{NaCl}$

3. Which of the following statements concerning differences between ionic and covalent compounds is correct?  
A) Covalent compounds contain 2 elements, and ionic compounds contain 3 or more elements.  
B) Ionic compounds contain oxygen, and covalent compounds do not.  
C) Covalent compounds form molecules, and ionic compounds do not.  
D) more than one correct response.  
E) no correct response.

4. Which of the following compounds would you expect to be ionic?  
A)  $\text{SF}_6$  B)  $\text{H}_2\text{O}$  C)  $\text{H}_2\text{O}_2$   
D)  $\text{NH}_3$  E)  $\text{CaO}$

5. Which chemical group is called the alkali metals?  
A) group 1 B) group 2 C) groups 3-12  
D) group 17 E) group 18

6. The number of protons in an atom of  $^{65}\text{Zn}$  is:  
A) 65 B) 35 C) 30 D) 1.

7. An atom of atomic number 53 and mass number 127 contains how many neutrons?  
A) 53 B) 74 C) 127 D) 180

8. Most of the elements on the periodic table are \_\_\_\_\_.  
A) gases B) nonmetals C) metalloids D) liquids E) metals

9. Which one of the following is NOT true about the alkali metals?  
A) They are low density solids at room temperature.  
B) They all readily form ions with a +1 charge.

# Physical Science Chemistry Test Answers Unit 2

**LP Steffe**



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