

Physical Science Chapter 2 Test**Completion**

Completion: write complete or statements.

1. Based on our model of matter because of how we _____.
2. Atoms, ions, molecules, and compounds are composition of _____.
3. The chemical symbol of oxygen for the element _____.
4. A substance that is made of atoms of same kind are kept bound together is called a _____.
5. Every chemical compound is unique and different from the _____ in nature.
6. The smallest particle of oxygen that behaves like oxygen is an oxygen _____.
7. The chemical formula for a chlorine molecule, which consists two chlorine atoms, is _____.
8. Each molecule of ammonia, NH_3 , contains _____ atoms of ions.
9. Each molecule of hydrochloric acid HCl contains _____ hydrogen atoms.
10. When two or more pure substances are mixed together, the result is a _____.
11. Two or more liquids that are different mix with each other are referred to as _____.
12. A mixture of dust and water is called a _____ mixture because the substances do not dissolve.
13. The kinetic theory states that the particles in matter are always in _____.
14. The kinetic theory states that, at the same _____, heavier particles move more slowly than lighter particles.
15. As gas molecules travel around each other, they spread in _____.
16. A solid holds its shape because its molecules are _____.
17. As heat is added to a solid substance, the atoms _____ and move apart.
18. When water _____, steam is released.
19. One way change of state is called _____, which is evaporation.
20. The surface of the liquid has several smaller air bubbles smaller because of _____.
21. Energy can be converted from one form to another, but it cannot be _____.
22. When you heat gasoline in a car's engine, you produce energy to move the car plus energy is the thermal _____.
23. When a piece of paper is burned, the amount of matter before is _____ the amount of matter afterward.

Physical Science Chapter 2 Test

J Dewey



Physical Science Chapter 2 Test:

Reviewing **Physical Science Chapter 2 Test**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Physical Science Chapter 2 Test**," an enthralling opus penned by a very acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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