Physical or Chemical Properties/Changes

Read each sta	itement and decide if it describes a physical or chemical property or change.	100
_PP_1.	Copper compounds are often blue or green in color.	
CP2	Copper objects often turn blue or green when exposed to air.	
CP3	Oxygen gas supports the combustion of fuel.	
	Metals, when pure, have a high luster (are shiny and bright).	
CP 5.	Some metals may become dull when exposed to air.	
AND DESCRIPTION OF THE PERSON NAMED IN COLUMN 1	Milk turns sour if left too long at room temperature.	
	Mercury's high density and liquid state at room temperature make it useful in	
thermometers.		
PPa	Metals are typically ductile and can be drawn into thin wires.	
PP s.	The hardness of diamonds enables them to be used on drill bits.	
CP10.	Bread dough increases in volume if it is allowed to "rise" before baking.	
	Tungsten's high melting point makes it useful for light bulb filaments.	
the state of the s	Metals are typically good conductors of heat and electricity.	****
and the second s		
13.	Two liquids are poured together and a solid substance is formed.	
CP14	When water is added to cold medicine, bubbles begin to form.	
Write three star changes. The	tements describing chemical changes and three statements describing physical statements above can give you ideas, but don't copy the statement. Each change int. (i.e. Do not cook everything)	
Chemical		
1. A log	burns on the fire	
2 Grand	Ipu decays in his grave	
	lysis branks up water into hydrogen a	- L
exy 3ª		
I-ILLASACINI	dissolves in water	
2 A shi	eld is cloven in two	-
3. Sodo	goes stat.	

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S Marginson

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