

Element	symbol	Atomic number	Electronic arrangement in different orbit					
			K	L	M	N	O	P
Hydrogen	H	1	1					
Helium	He	2	2					
Lithium	Li	3	2	1				
Beryllium	Be	4	2	2				
Boron	B	5	2	3				
Carbon	C	6	2	4				
Nitrogen	N	7	2	5				
Oxygen	O	8	2	6				
Fluorine	F	9	2	7				
Neon	Ne	10	2	8				
Sodium	Na	11	2	8	1			
Magnesium	Mg	12	2	8	2			
Aluminium	Al	13	2	8	3			
Silicon	Si	14	2	8	4			
Phosphorus	P	15	2	8	5			
Sulphur	S	16	2	8	6			
Chlorine	Cl	17	2	8	7			
Argon	Ar	18	2	8	8			
Potassium	K	19	2	8	8	1		
Calcium	Ca	20	2	8	8	2		

● $c = \nu\lambda$

[c = velocity of light in free space,
 ν = frequency, λ = wavelength]

● $\bar{\nu} = \frac{1}{\lambda} = \frac{\nu}{c}$

[ν = wave number]

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C Cleary

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Chemical

Interactions, 2005 The Development of Modern Chemistry Aaron J. Ihde, 1984-01-01 From ancient Greek theory to the explosive discoveries of the 20th century this authoritative history shows how major chemists their discoveries and political economic and social developments transformed chemistry into a modern science 209 illustrations 14 tables Bibliographies Indices Appendices

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Board University of Adelaide. Public Examinations Board, 1959 The Manuals include information on syllabus regulations copies of examination papers and notes by examiners They also include pass lists *Soviet Marxism and Natural Science* David Joravsky, 2013-04-15 Originally published in 1961 Russian Marxist philosophy of science originated among men and women who gave their whole lives to rebellion against established authority The original tension within Marxist philosophy between positivism and metaphysics was repressed but not resolved in this first phase of Soviet Marxism In this volume the

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